

GROUND WATER RESOURCES



Marion County

Open File Report 80-63 WRD

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GROUND-WATER RESOURCES OF MARION COUNTY

Introduction

Approximately 86% of the residents of Marion County rely on ground water as the source of their drinking water. It is estimated that the use of ground water in the county currently approaches 1.6 billion gallons per year. For comparison, this amount would provide each resident with 136 gallons of water a day during the year. Actually, few if any households use this much water, and the rather large annual per capita use reflects the greater water requirements of the county's industries, agribusinesses and municipalities.

The users of ground water in the county draw their supplies from several different geologic sources. Several factors must be considered in determining the availability of ground water and the adequacy of a supply source:

distribution - having water where it is needed,

accessibility - affects the costs for drilling wells and pumping water,

yield - relates to the magnitude of the supply that can be sustained,

quality - determines for what purposes the water can be used.

In terms of these factors, there are few locations in Marion County where the availability of ground water is not limited to some degree. The most common limitation is poor water quality, that is, highly mineralized ground water. Secondary limitations are generally related to poor distribution, small yields from some sources, and poor accessibility due to the great depths to adequate sources.

Occurrence of Ground Water in Marion County

The occurrence of ground water is influenced by geology -- the position and thickness of the rock units, their ability to store and transmit water, and their physical and chemical make-up. Geologic units that store and transmit water and yield appreciable amounts to wells are called aquifers. The best aquifers are usually composed of unconsolidated sand and gravel, porous sandstone, and porous or fractured limestone and dolostone. Other units with materials such as clay and silt, shale, siltstone, and mudstone yield little or no water to wells. These impermeable units are called aquicludes or aquitards and commonly separate one aquifer unit from another.

In Marion County there are three principal sources from which users obtain water supplies; the loose, unconsolidated materials near the land surface which comprise surficial aquifers; below there are three major rock aquifers -- the Mississippian, the Devonian, and the Cambro-Ordovician. Each of the aquifers has its own geologic, hydrologic, and water quality characteristics which determine its relative suitability as a water supply source.

Surficial Aquifers

Unconsolidated deposits above the bedrock are comprised of mixtures of clay, silt, sand, gravel, and assorted boulders. The water-yielding potential of these deposits is greatest for units composed mostly of sand and/or gravel. The three types of surficial aquifers used in the county are: alluvial aquifers, the drift aquifer and buried channels.

The alluvial aquifers consist mainly of the sand and gravel transported and deposited by modern streams along floodplains or on terraces in major stream valleys. Alluvial deposits tend to be shallow, generally less than 50-60 feet and thus may be easily contaminated by infiltrating surface waters.

The drift aquifer is the thick layer of clay to boulder size material deposited on top of the bedrock by glaciers which invaded the county at least twice in the last two million years. The composition of the glacial drift varies considerably and in many places does not yield much water. There are, however, lenses or beds of sand and gravel within the drift that are thick and widespread enough to serve as dependable water sources. These are often difficult to locate because they are irregular in shape and are located between other drift deposits. Usually one or two sand layers can be found in most places that will yield enough water to supply a domestic well.

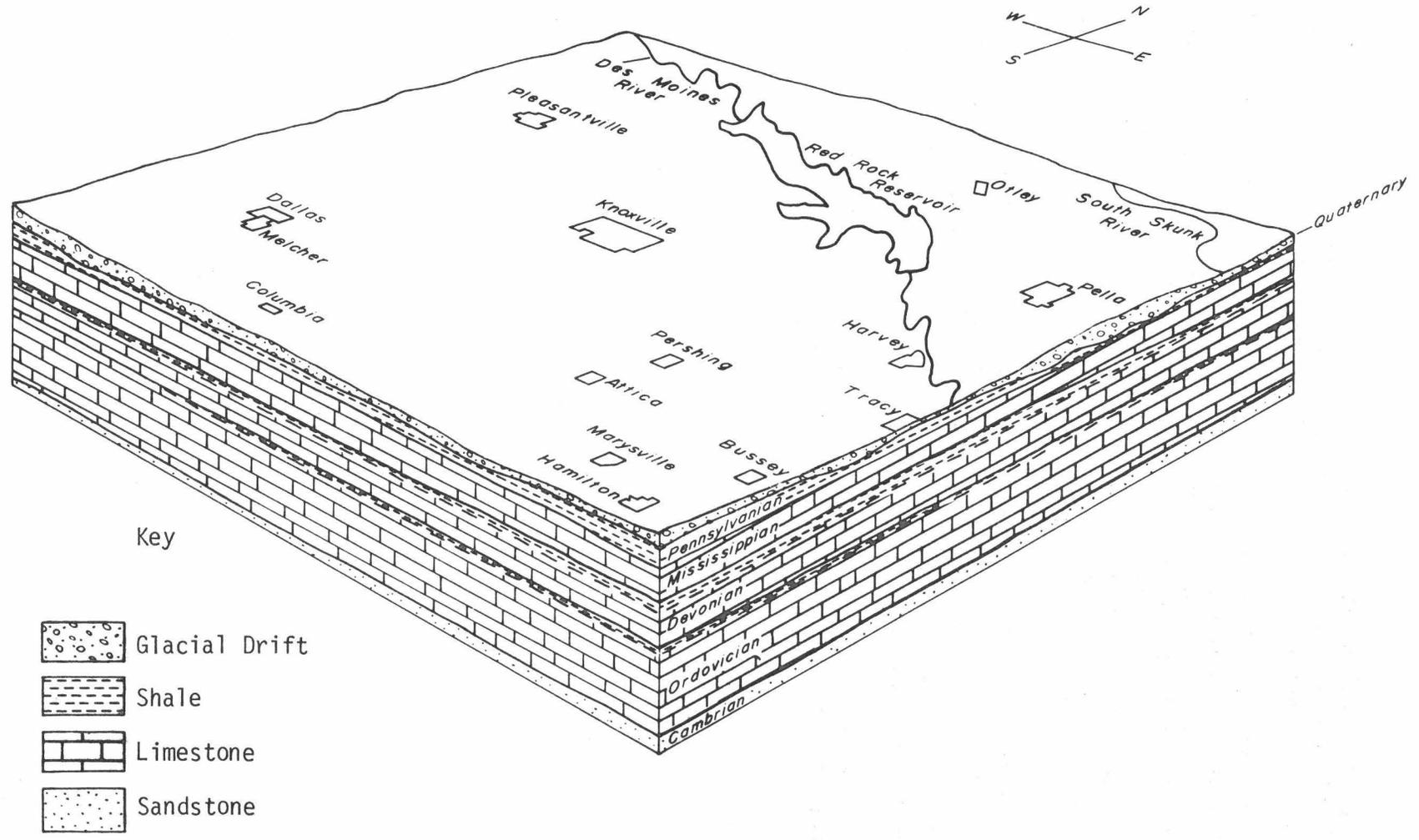
The buried channel aquifer consists of alluvial fill in stream valleys that existed before the glacial period. The valleys were overridden by glaciers and are now buried beneath the drift or more recent alluvial deposits.

The distribution, yields and water quality characteristics for the surficial aquifers are summarized in Figures 2 and 9 and Table 3. An indication of accessibility can be obtained by comparing the elevations of the top (the land surface) and the bottom (the bedrock surface) of the surficial deposits from Figure 4 and 5. The thickness of the glacial drift and the depth of the buried channels are determined by subtracting the elevations at selected locations.

Rock Aquifers

Below the surficial materials is a thick sequence of layered rocks formed from deposits of rivers and shallow seas that have covered the state within the last 600 million years. The geologic map (Figure 3) shows the geologic units which form the upper surface of this rock sequence. Over most of the county rocks of Pennsylvanian age lie directly below the glacial drift. These rocks in Marion County are primarily shales, thin coal beds, limestones, and sandstones. Because shales predominate, the Pennsylvanian sequence acts as an aquiclude across the county and only locally can water be produced from it. These localities are in the western half of the county and the water is produced from sandstone units of the Cherokee Group.

Underlying the Pennsylvanian aquiclude are older rocks, portions of which form the major rock aquifers in Marion County. These rocks and their water-bearing characteristics are shown in Table 1.



Key

-  Glacial Drift
-  Shale
-  Limestone
-  Sandstone

Figure 1

BLOCK DIAGRAM SHOWING THE GEOLOGY OF MARION COUNTY

Examples of the rock units encountered in several wells at various locations in Marion County are indexed and illustrated in Figures 7 and 8. The geologic unit that supplies ground water and the rate of yield are shown for each well.

The relative accessibility of ground water in rock aquifers depends on the depth to the aquifer. The deeper a well must be, the greater the cost for well construction and pumping. The depths to and thicknesses of units at specific sites will vary somewhat because of irregularities in the elevation of the land surface and in the elevation of the underlying rock units. Estimates of depths and thicknesses can be made by comparing Figure 4 with the maps of aquifer elevations in Figures 11, 12 and 13. The range in depth below land surface to the top of the county's principal bedrock aquifers is given for each township in Figure 6.

A second factor affecting ground water accessibility is the level to which the water will rise in a well (the static water level). Throughout the county water in the rock aquifers is under artesian pressure and rises in the well once the aquifer is penetrated. This can reduce the cost of pumping. Average static water levels for Marion County wells are shown in Figures 11, 12 and 13.

Average rates of yield and water quality characteristics for each of the aquifers are summarized in the maps in Figures 11, 12 and 13 and Table 14.

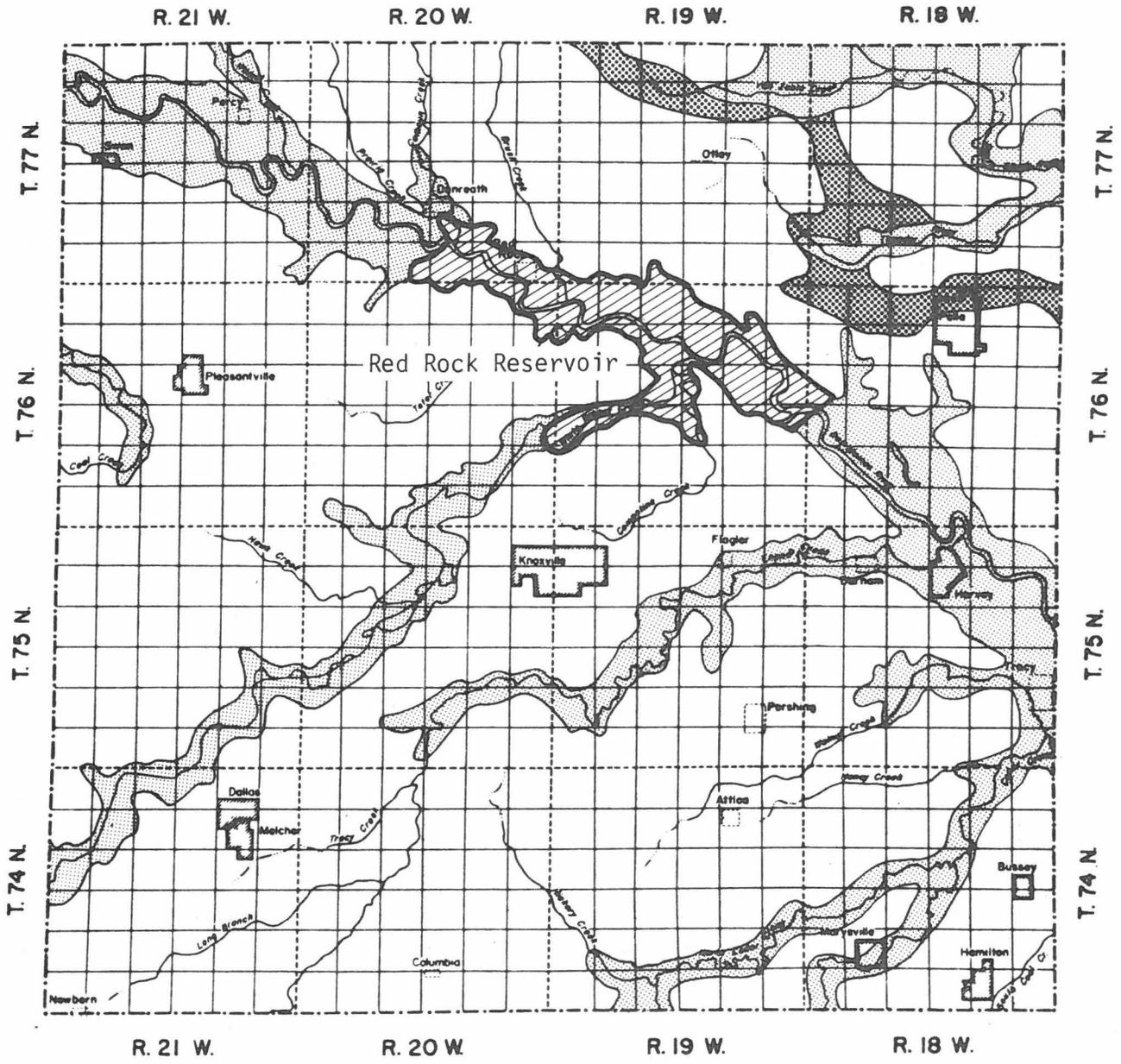
Table 1

GEOLOGIC AND HYDROGEOLOGIC UNITS IN MARION COUNTY

Age	Rock Unit	Description	Thickness Range	Hydrogeologic Unit	Water-Bearing Characteristics
Quaternary	Alluvium	Sand, gravel, silt and clay	0-400 (feet)	Surficial aquifer	Fair to large yields (10 to 500 gpm)
	Glacial drift (undifferentiated)	Predominantly till containing scattered irregular bodies of sand and gravel			Low yields (less than 10 gpm)
	Buried channel	Sand, gravel, silt and clay			Small to large yields
Pennsylvanian	Marmaton Group	Alternating shale and limestone; thin coal and sandstone	0-500	Aquiclude	Low yields only from limestone and sandstone
	Cherokee Group	Shale, clay, siltstone, sandstone and coal beds, mostly thin			
Mississippian	Meramec Series	Sandy limestone	200-550	Mississippian aquifer	Fair to low yields
	Osage Series	Limestone and dolostone, cherty; shale			
	Kinderhook Series	Limestone, oolitic, and dolostone, cherty			
Devonian	Maple Mill Shale Sheffield Formation Lime Creek Formation	Shale, limestone in lower	100-250	Devonian aquiclude	Does not yield water
	Cedar Valley Limestone Wapsipinicon Formation	Limestone and dolostone, contains evaporites (gypsum) in southern half of Iowa	400-450	Devonian aquifer*	Fair to low yields
Silurian	Undifferentiated	Dolostone	0-40	Silurian aquifer	Low yields
Ordovician	Maquoketa Formation	Shale and dolostone	925-1000	Maquoketa aquiclude	Does not yield water
	Galena Formation	Dolostone and chert		Minor aquifer	Low yields
	Decorah Formation-Platteville Formation	Limestone, dolostone and thin shale, includes sandstone in SE Iowa		Aquiclude	Does not yield water
	St. Peter Sandstone	Sandstone		Cambro-Ordovician aquifer	Fair yields
	Prairie du Chien Formation	Dolostone, sandy and cherty			High yields (over 500 gpm)
Cambrian	Jordan Sandstone	Sandstone	20-50+	Aquitard	Low yields
	St. Lawrence Formation	Dolostone			
	Franconia Sandstone	Sandstone and shale			
	Dresbach Group	Sandstone	Dresbach aquifer*		
Precambrian	Undifferentiated	Coarse sandstone: crystalline rocks		Base of ground water reservoir	Not known to yield water

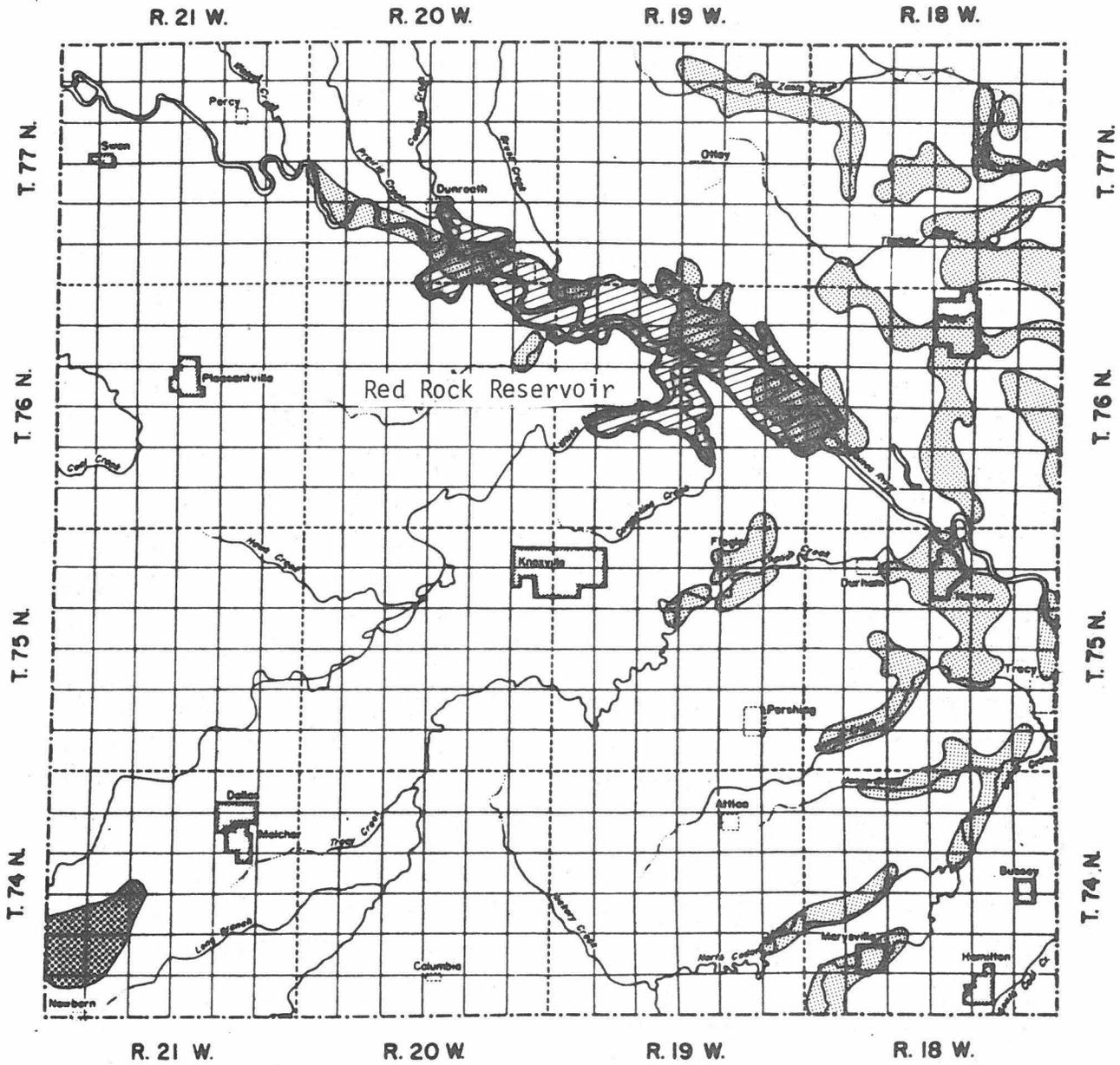
*not significant in Marion County owing to highly mineralized water contained.

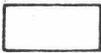
Figure 2
SURFICIAL MATERIALS



-  Alluvium
-  Glacial Drift
-  Buried Channels

Figure 3
GEOLOGIC MAP

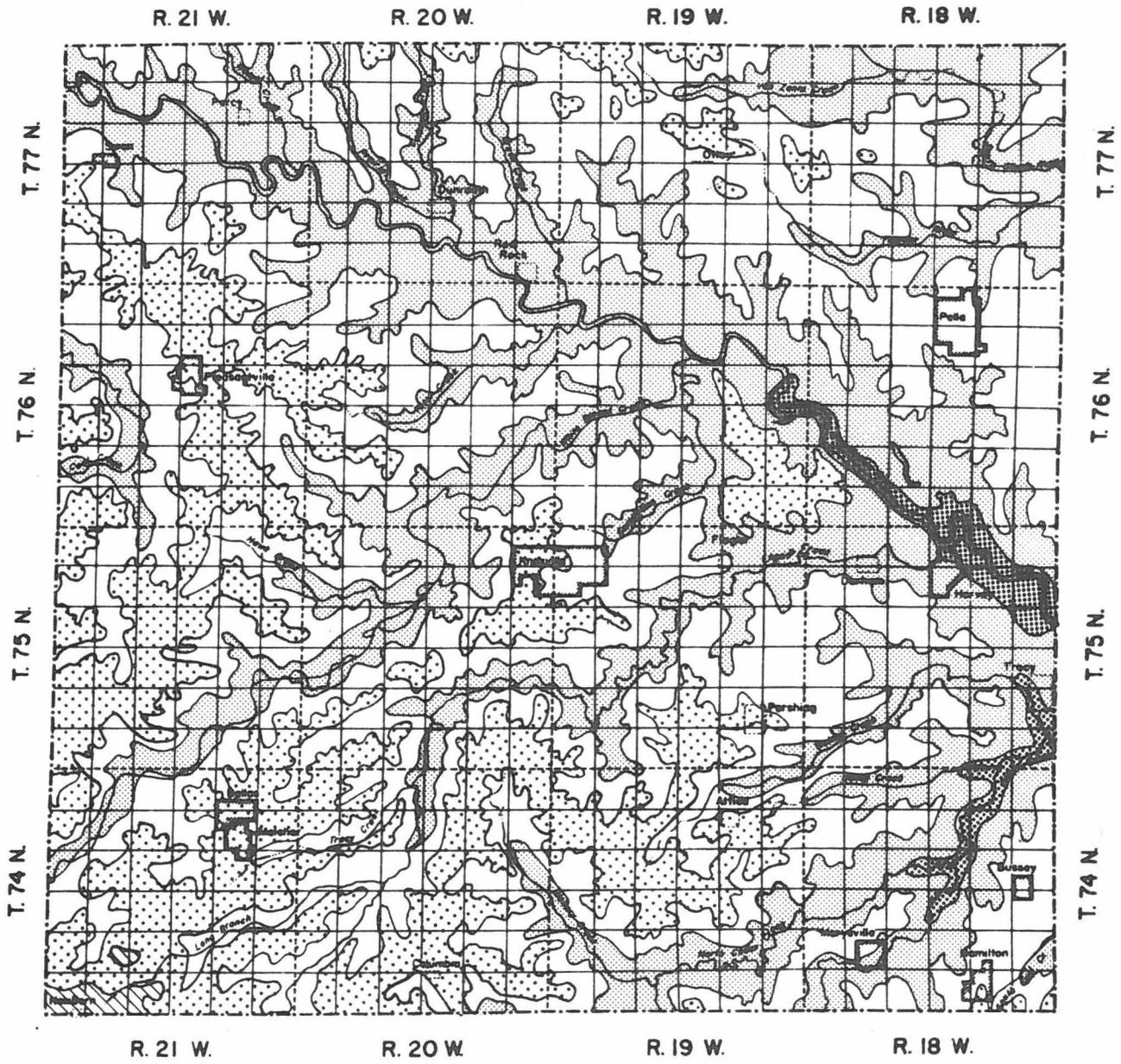


 Pennsylvanian
 Marmaton Group
 Pennsylvanian
 Cherokee Group

 Mississippian

Figure 4

ELEVATION OF LAND SURFACE IN FEET ABOVE MEAN SEA LEVEL



 Below 700
 700-800
 800-900

 900-1000
 1000-1100

Figure 5

ELEVATION OF BEDROCK SURFACE IN FEET ABOVE MEAN SEA LEVEL

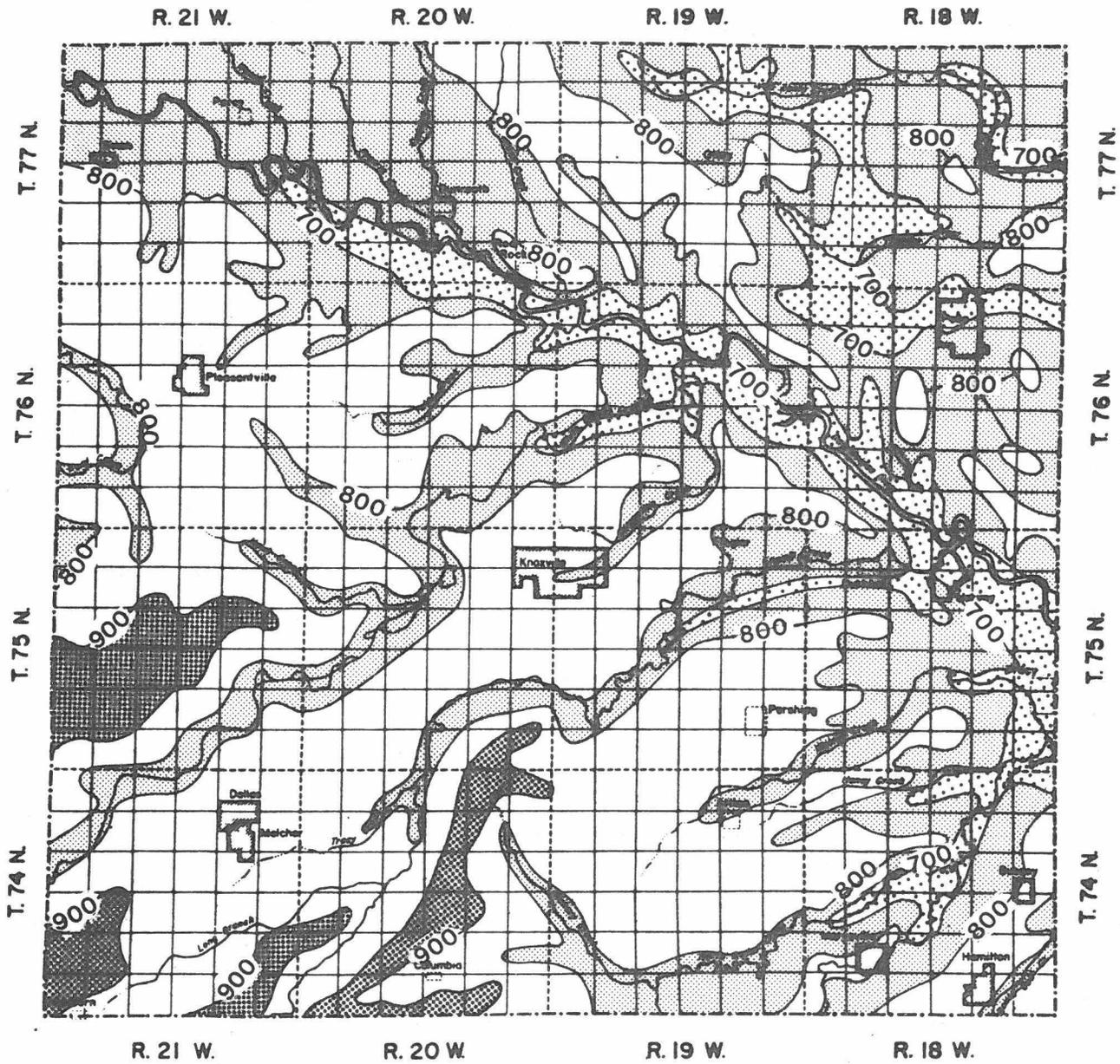


Figure 6

RANGE IN DEPTH TO MARION COUNTY'S PRINCIPAL ROCK AQUIFERS

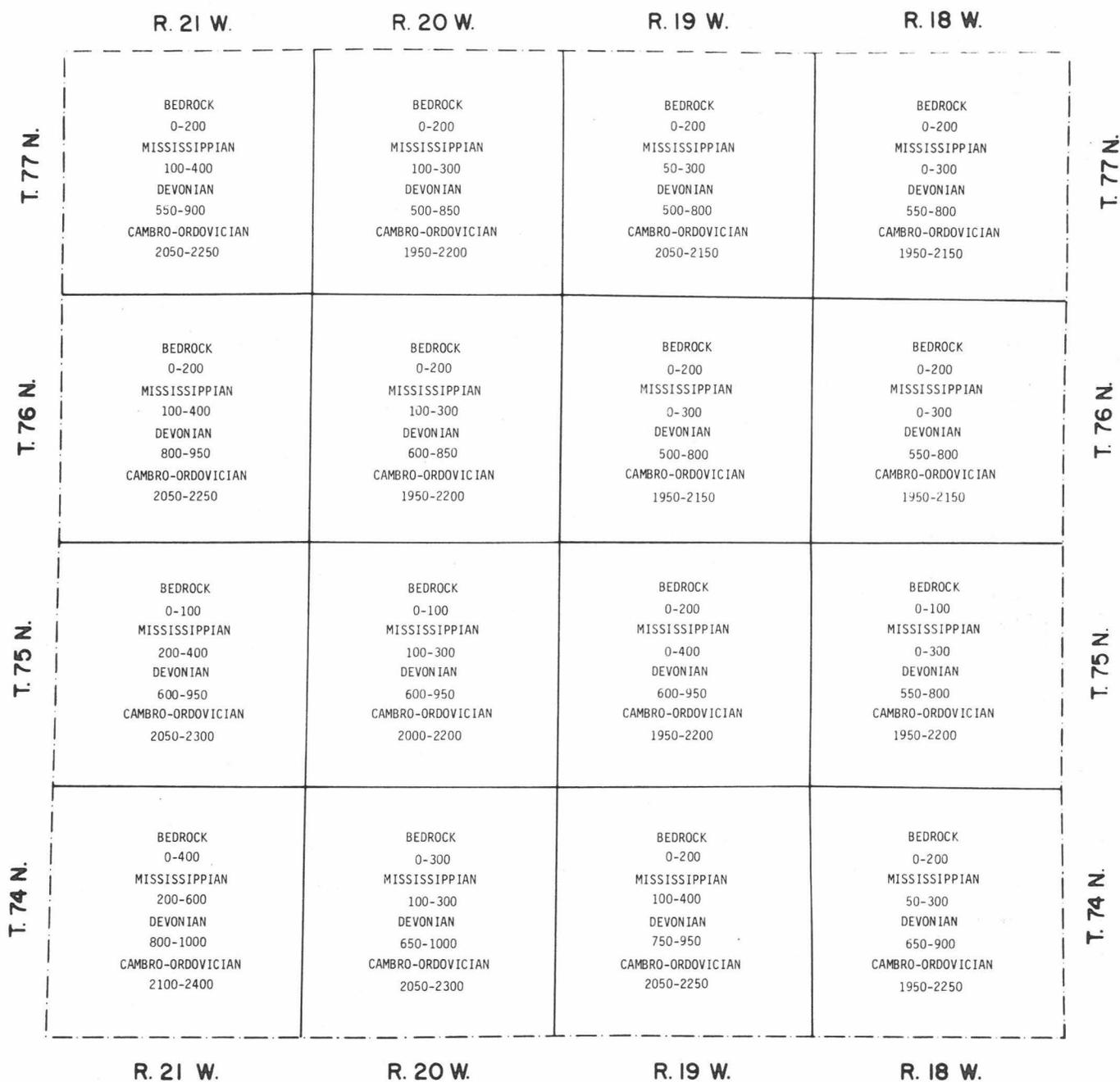


Figure 7

INDEX MAPS FOR TYPICAL WELLS IN MARION COUNTY

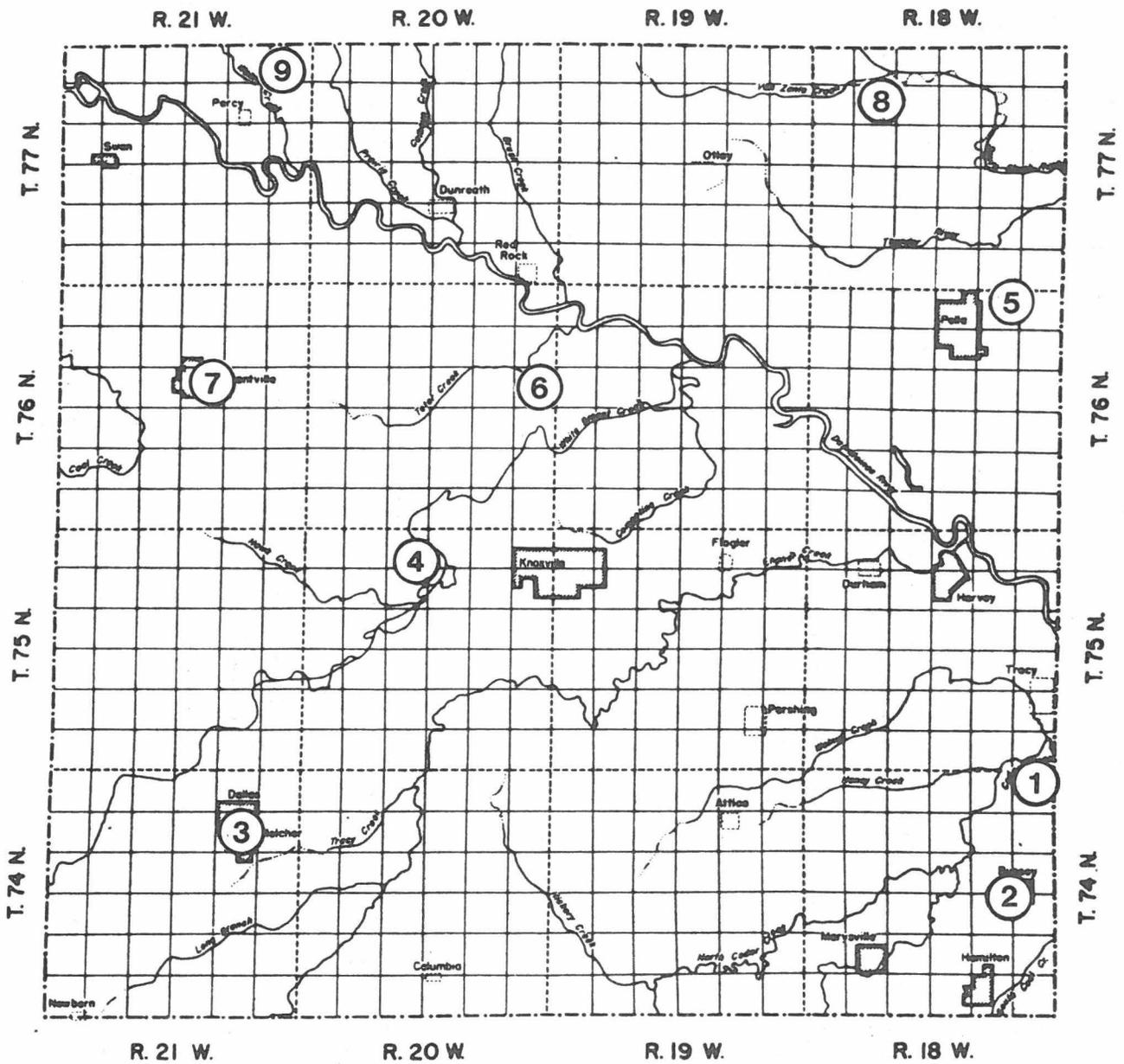


Figure 8

TYPICAL WELLS IN MARION COUNTY

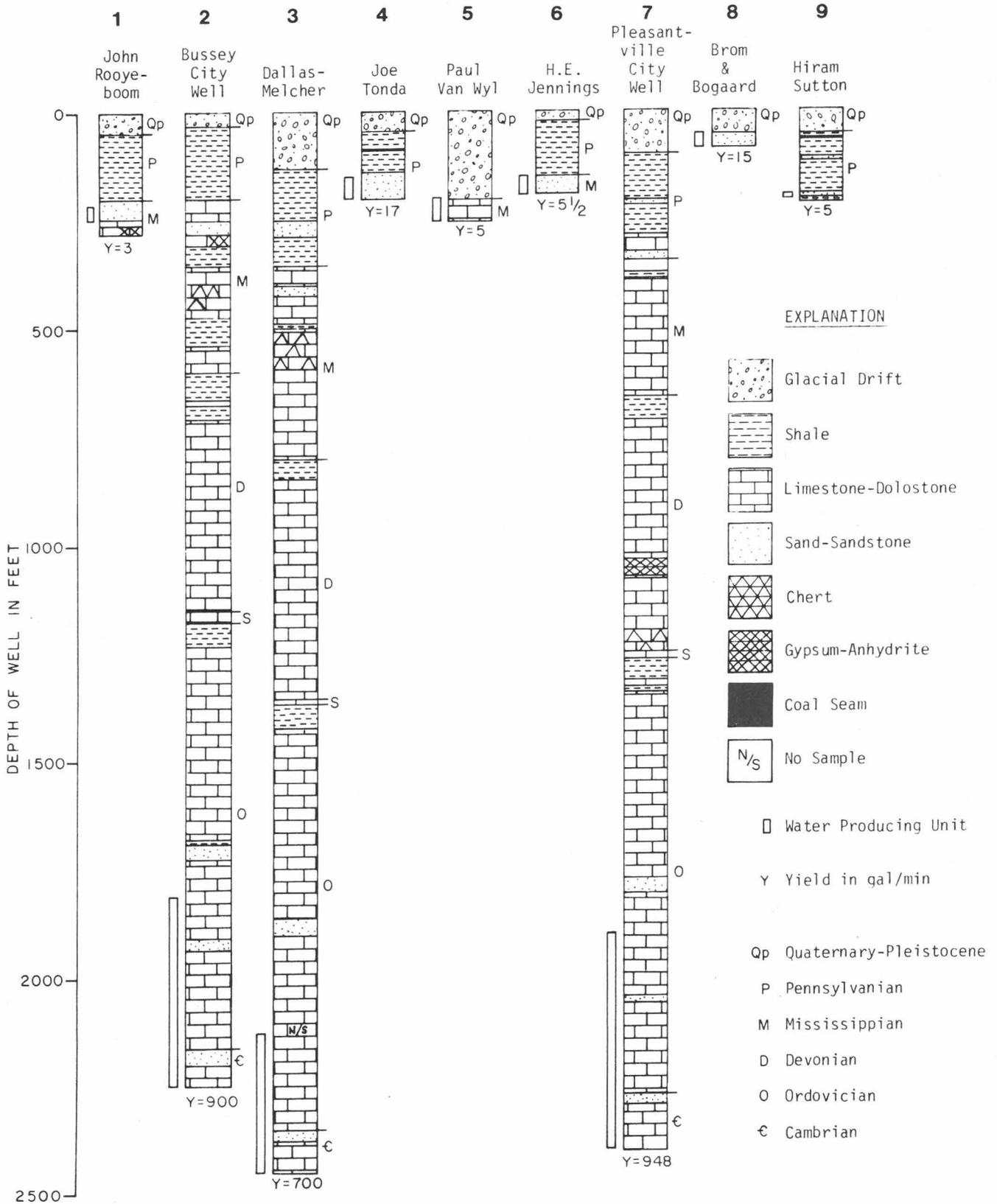


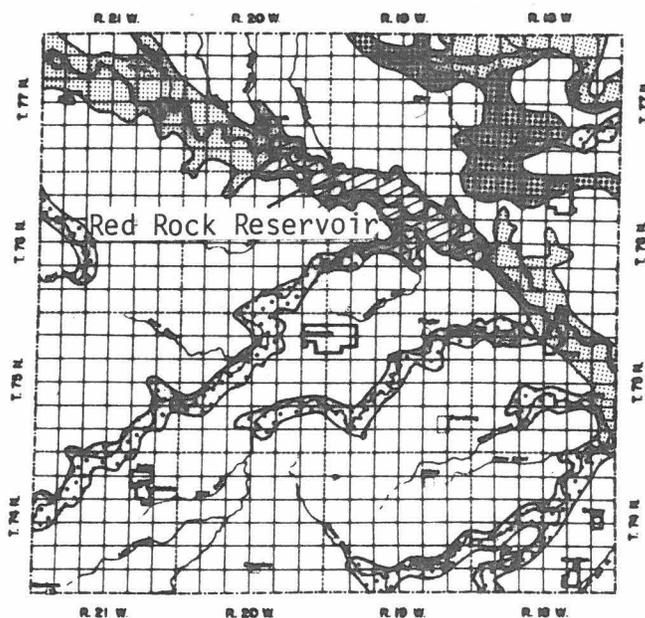
Figure 9

SURFICIAL AQUIFERS

Water Levels

Water levels in the surficial aquifers are difficult to analyze, water rises to different levels in wells drilled into alluvial, buried-channel, and drift aquifers. The water table in the drift aquifer generally slopes from high land areas toward the streams and, changes noticeably throughout the year. Levels in drift and buried-channel aquifers respond rapidly to recharge from precipitation. Water levels in the alluvial aquifer fluctuate somewhat in the same way as those in the drift and buried-channel aquifers; however, the main influence on the alluvial aquifer is the stage (level) of the associated streams. Water levels will be high during periods of high stream stage and low during the low-stage periods.

Water levels in the drift aquifer commonly are from 10 to 50 feet below the land surface, and those in the buried-channel aquifers have been reported to be as low as 175 feet below the land surface. The water levels in alluvial wells are from 4 to 20 feet below the flood-plain surface and the depth to the water surface will be accordingly deeper in wells located on terrace surfaces.



Water yields to wells in gallons per minute

Alluvial aquifers

Drift aquifer

10-25 (35)*

1-5 (10)

25-50 (75)

Buried channel and drift aquifers

150-500 (1000)

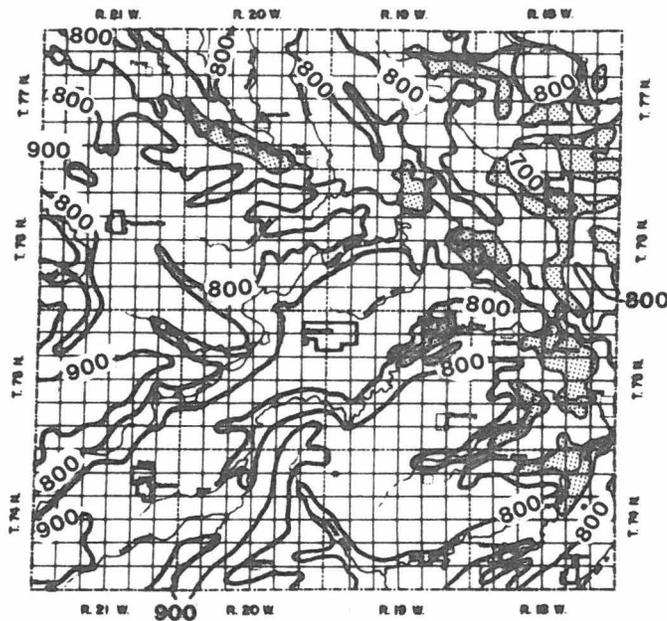
12-25 (60)

*(35) number is maximum yield that is occasionally available but probably not on a sustained basis.

Figure 10

Pennsylvanian Aquiclude

The Pennsylvanian Aquiclude generally underlies the whole county and is thickest in the western part of the county. Pennsylvanian strata consist of a succession of predominantly shale beds of the Marmaton Group and Cherokee Group, with occasional thin layers of sandstone (in the Cherokee) and limestone (in the Marmaton), which locally functions as aquifers. Coal is present both in the Marmaton and Cherokee groups and is present in thin beds or lenses which range in thickness from a few inches up to 5 feet.



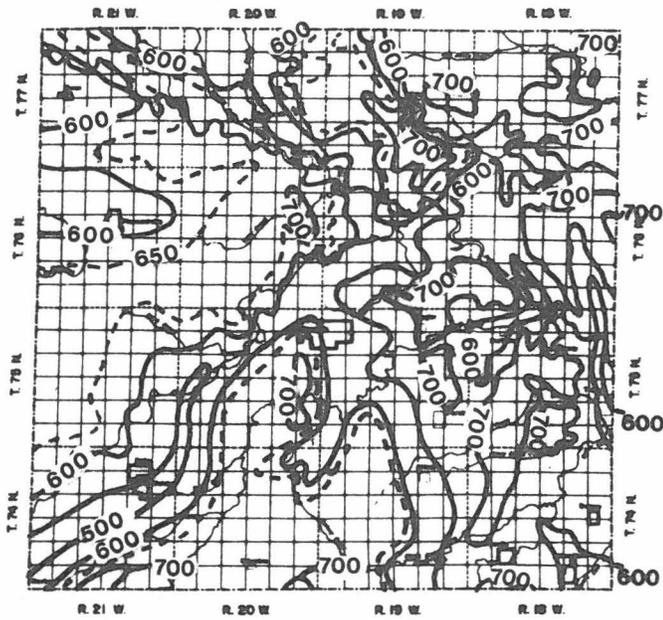
Elevation of the top of the Pennsylvanian Aquiclude in feet above mean sea level



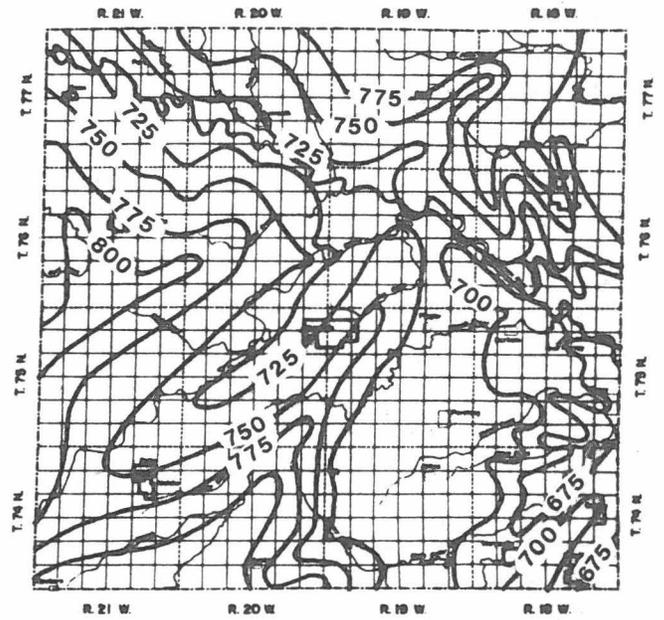
Pennsylvanian Aquiclude not present

Figure 11

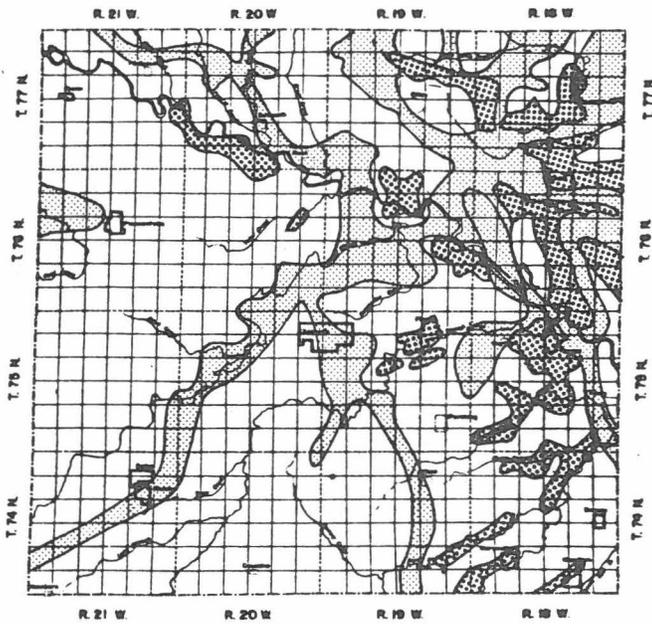
MISSISSIPPIAN AQUIFER



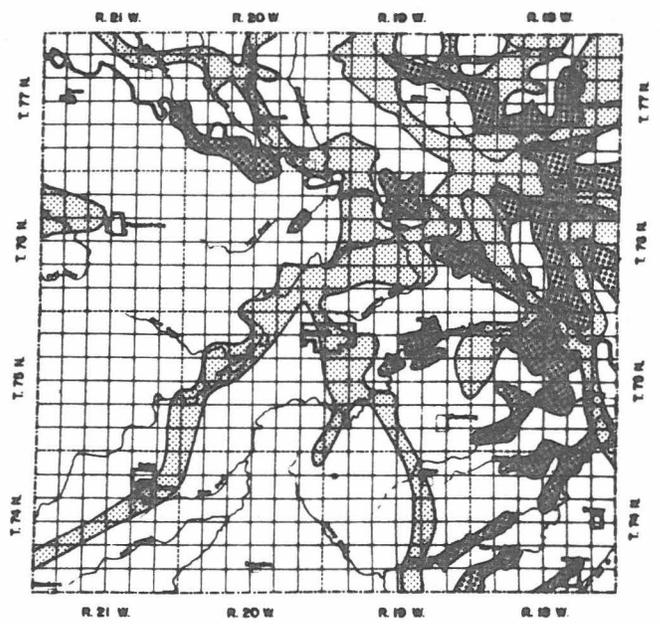
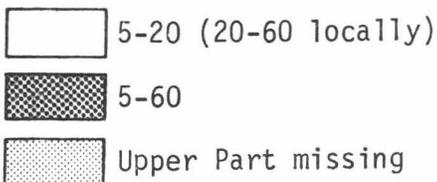
Elevation of Mississippian Aquifer in feet above mean sea level



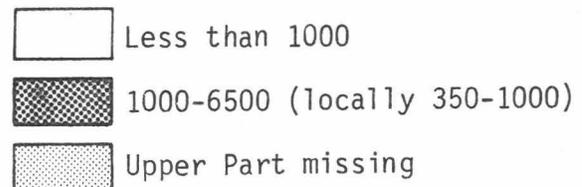
Water levels in wells in feet above mean sea level



Water yields to wells in gallons per minute



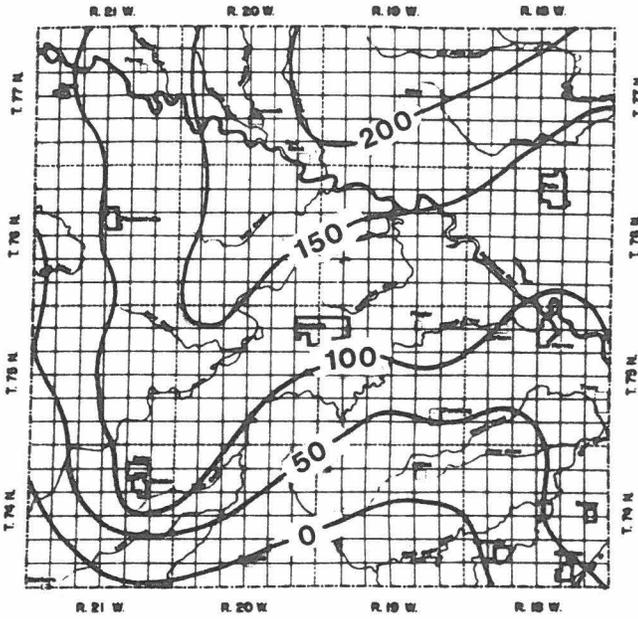
Dissolved solids content in milligrams per liter (mg/l)*



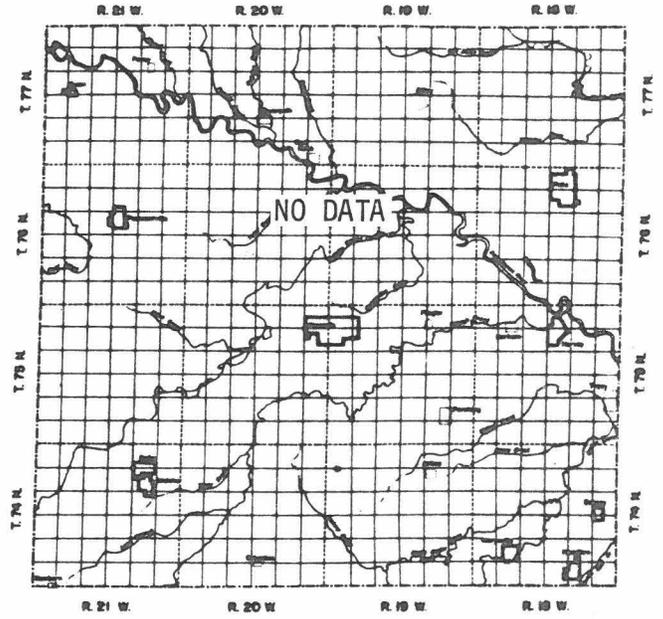
* Other water quality data in Table 4

Figure 12

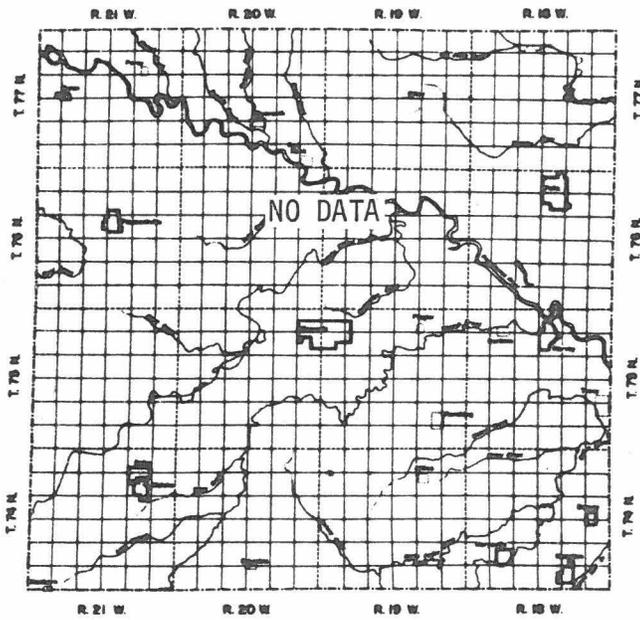
DEVONIAN AQUIFER



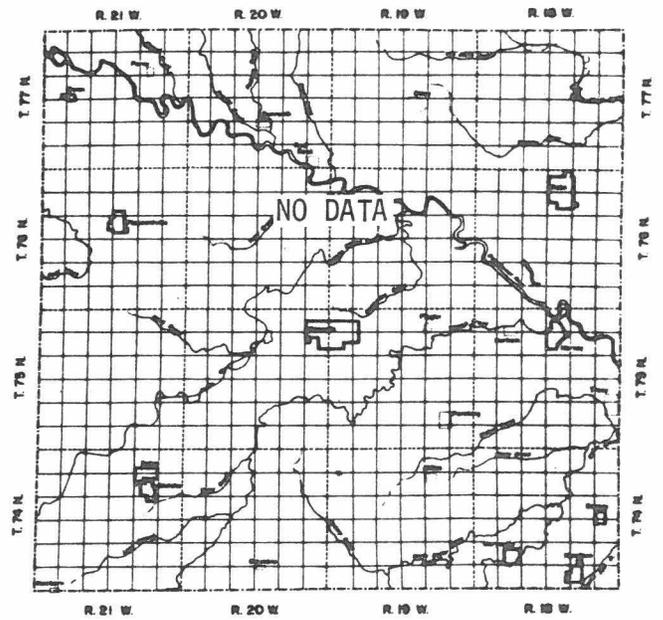
Elevation of Devonian Aquifer in feet above mean sea level



Water levels in wells in feet above mean sea level



Water yields to wells in gallons per minute

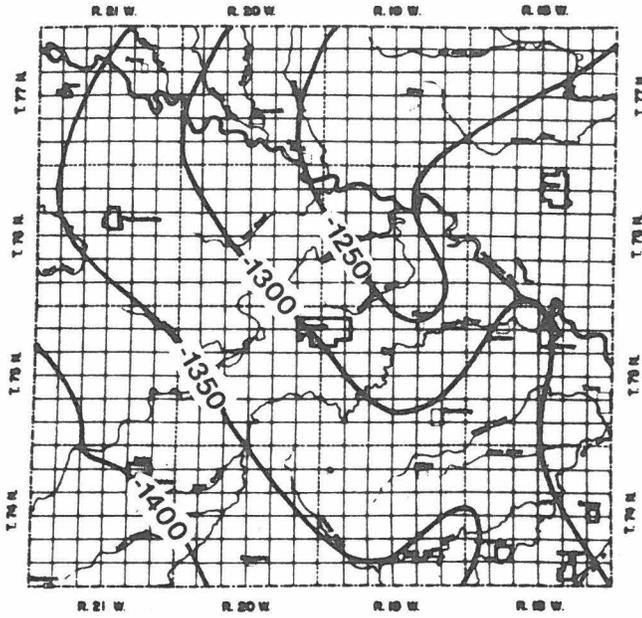


Dissolved solids content in milligrams per liter (mg/l)*

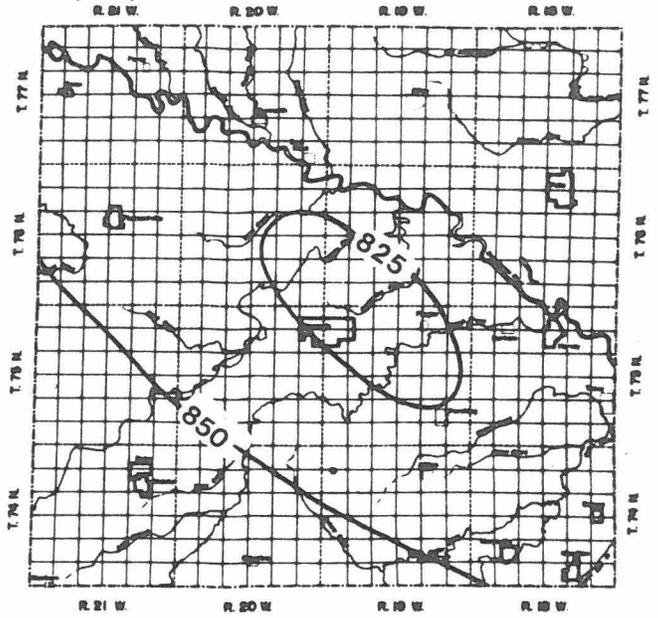
*Other water quality data in Table 4

Figure 13

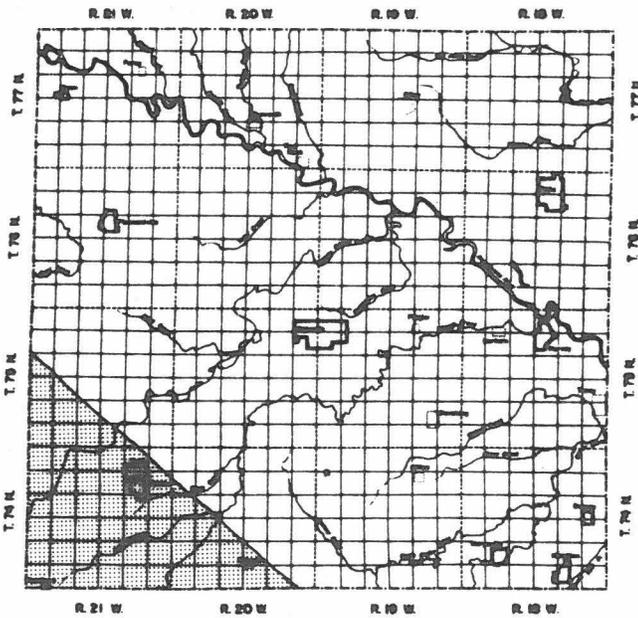
CAMBRO-ORDOVICIAN (JORDAN) AQUIFER



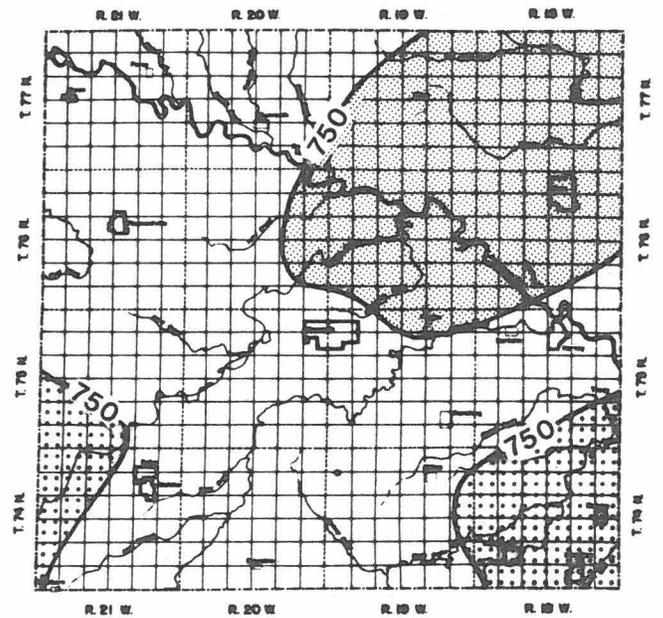
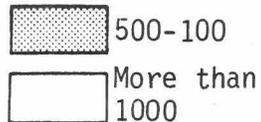
Elevation of Jordan Aquifer in feet above sea level



Water levels in wells in feet above mean sea level



Water yields of wells in gallons per minute



Dissolved solids content in milligrams per liter (mg/l)*



*Other water quality data in Table 4

Table 2

SIGNIFICANCE OF MINERAL CONSTITUENTS AND PHYSICAL PROPERTIES OF WATER

Constituent or Property	Maximum Recommended Concentration	Significance
Iron (Fe).....	0.3 mg/l.....	Objectional as it causes red and brown staining of clothing and porcelain. High concentrations affect the color and taste of beverages.
Manganese (Mn).....	0.05 mg/l.....	Objectionable for the same reasons as iron. When both iron and manganese are present, it is recommended that the total concentration not exceed 0.3 mg/l.
Calcium (Ca) and Magnesium (Mg).....		Principal causes for hardness and scale-forming properties of water. They reduce the lathering ability of soap.
Sodium (Na) and Potassium (K).....		Impart a salty or brackish taste when combined with chloride. Sodium salts cause foaming in boilers.
Sulfate (SO ₄).....	250 mg/l.....	Commonly has a laxative effect when the concentration is 600 to 1,000 mg/l, particularly when combined with magnesium or sodium. The effect is much less when combined with calcium. This laxative effect is commonly noted by newcomers, but they become acclimated to the water in a short time. The effect is noticeable in almost all persons when concentrations exceed 750 mg/l. Sulfate combined with calcium forms a hard scale in boilers and water heaters.
Chloride (Cl).....	250 mg/l.....	Large amounts combined with sodium impart a salty taste.
Fluoride (F).....	2.0 mg/l.....	In central Iowa, concentrations of 0.8 to 1.3 mg/l are considered to play a part in the reduction of tooth decay. However, concentrations over 2.0 mg/l will cause the mottling of the enamel of children's teeth.
Nitrate (NO ₃).....	45 mg/l.....	Waters with high nitrate content should not be used for infant feeding as it may cause methemoglobinemia or cyanosis. High concentrations suggest organic pollution from sewage, decayed organic matter, nitrate in the soil, or chemical fertilizer.
Dissolved solids.....	500 mg/l.....	This refers to all of the material in water that is in solution. It affects the chemical and physical properties of water for many uses. Amounts over 2,000 mg/l will have a laxative effect on most persons. Amounts up to 1,000 mg/l are generally considered acceptable for drinking purposes if no other water is available.
Hardness (as CaCO ₃)..		This affects the lathering ability of soap. It is generally produced by calcium and magnesium. Hardness is expressed in milligrams per liter equivalent to CaCO ₃ as if all the hardness were caused by this compound. Water becomes objectionable for domestic use when the hardness is above 100 mg/l; however, it can be treated readily by softening.
Temperature.....		Affects the desirability and economy of water use, especially for industrial cooling and air conditioning. Most users want a water with a low and constant temperature.

To the user, the quality of ground water is as important as the amount of water that an aquifer will yield. As ground water moves through soil and rock materials, it dissolves some of the minerals which, in turn, affect water quality. In addition to mineral content, bacterial and chemical contamination may be introduced through poorly constructed wells and seepage from other pollution sources.

Recommended standards for common water constituents are described in the table above. These are rationally accepted as guidelines for acceptable drinking water supplies. Limits for uses other than drinking often differ from these. For instance, water that is unacceptable for drinking and household use may be completely satisfactory for industrial cooling.

From analyses of ground water averages (A) and ranges (R) of values in milligrams per liter (mg/l) for several constituents are summarized in Tables 3 and 4 for the surficial and bedrock aquifers in Marion County. Recommended concentrations for some constituents are often exceeded without obvious ill effects, although the water may be unpalatable. Water quality analyses for individual wells should be obtained to determine if concentrations of constituents that affect health are exceeded.

Table 3

CHEMICAL CHARACTER OF GROUND WATER

Average (A) and range (R)	Dissolved solids	Hardness (as CaCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Sodium (Na)	Iron (Fe)	Manganese (Mn)
<u>Alluvial aquifer</u>									
A	417	288	105	13	0.3	4.2	20.2	9.4	1.5
R	165-1000	124-762	3-350	0.5-180	0.1-0.45	0.1-17	4.1-85	0.04-51	0.05-17
<u>Shallow drift aquifer</u>									
A	736	480	177	37	0.3	81	68	1.1	0.13
R	220-2840	153-1710	12-1470	0.5-200	0.2-0.8	0.1-570	7.3-710	0.02-30	0.05-1.9
<u>Intermediate drift aquifer</u>									
A	1030	569	397	9	0.5	6	108	5	0.09
R	261-2726	150-1518	7-1520	0.5-49	0.2-1.0	0-44	17-368	0.04-24	0.05-0.37
<u>Deep drift and Buried-channel aquifers</u>									
A	2346	868	1254	30	0.6	6.7	334	3.4	0.24
R	383-3657	140-1640	42-1990	3-110	0.1-2.0	0-82	54-568	0-18	0-1.4

The alluvial aquifers yield the least mineralized water of all ground water sources in south central Iowa. In the alluvial aquifers, manganese and iron concentrations are well above recommended standards, but all other constituents are well below. Water temperatures average 55°F (13°C) and the range of these temperatures is from 46°F to 60°F (8°C to 16°C).

In the shallow drift aquifers, the water is hard and contains undesirable concentrations of iron, sulfate, nitrate, chloride and dissolved solids. These high concentrations of nitrate, chloride and dissolved solids are generally due to contamination by direct runoff into the well and to infiltration to barnyard wastes, and cannot be overall accepted as representative of water under natural conditions in the shallow drifter aquifers, because they are locally contaminated. The water in these shallow drift aquifers is usually accepted for most purposes if wells are constructed properly and located a suitable distance from sources of contamination. Nitrate content should be checked carefully in these wells, and any water supply containing over 45 mg/l should not be used for infant feeding. Water temperatures average 54°F (12°C) and the range of these temperatures is from 50°F to 60°F (10°C to 16°C).

In the intermediate drift aquifer, water is more mineralized than the shallow drift aquifer, with iron concentrations high and nitrate low. The fluoride content, hardness and temperature are similar to the shallow drift aquifer.

In the deep drift and buried-channel aquifers, the water is highly mineralized and contains high concentrations of dissolved solids, sulfate and iron. Water temperatures range between 54°F and 57°F (12°C to 14°C).

Table 4

CHEMICAL CHARACTER OF GROUND WATER

Bedrock Aquifers

Average (A) and range (R)	Dissolved solids	Hardness (as CaCO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluoride (F)	Nitrate (NO ₃)	Sodium (Na)	Iron (Fe)	Manganese (Mn)
<u>Pennsylvanian (Cherokee Group) Aquifers</u>									
A	4531	869	1088	97	1.4	3.5	536	3.2	0.15
R	251-7092	44-1559	22-4139	0.5-780	0.2-4.0	0-50	7-2180	0.1-22	0-4
<u>Mississippian Aquifer</u>									
Upper Part of Mississippian aquifer in Marion County where aquifer is bedrock surface									
A	674	449	217	48	0.24	*	43	2.2	0.12
R	422-1380	108-952	27-563	2-32	0.2-0.4	0.1-4.3	8.8-240	0.2-9.8	0.05-0.25
Upper Part of Mississippian aquifer in Marion County where overlain by Pennsylvanian rocks									
A	2826	862	1493	109	1.3	1.6	492	6.6	0.06
R	368-6500	124-1660	49-3480	1-492	0-2.4	0.1-7.1	6.8-1480	0.1-45	0-0.12
Lower part of Mississippian aquifer in Marion County where overlain by upper part of aquifer									
A	3320	1004	1856	89	0.93	15.4	626	4.1	0.23
R	1189-4990	731-1159	582-3065	9-230	0-2.8	0.04-32	28-1091	0.18-25	0-1.3
<u>Cambrian-Ordovician aquifer</u>									
A	1098	370	397	150	2.3	1.2	226	2.4	0.05
R	614	246-1100	190-930	29-620	1.2-3.2	0.08-5.5	100-520	0.04-10	0.01-0.10

The water contained in the rock aquifers beneath Marion County all carry significant dissolved mineral concentrations. Among these, as can be noted in Table 4, is sulfate. Sulfate bearing minerals are found associated in the shales of the Pennsylvanian throughout the county and with the limestones in the lower part of the Mississippian aquifer in the eastern and southeastern parts of the county. Therefore, water from the Pennsylvanian is uniformly of poor quality, as is generally the case for the lower part of the Mississippian aquifer, and where the Mississippian aquifer (upper or lower part) is overlain by the Pennsylvanian. Water from the Devonian aquifer, based on data from adjacent counties, can be considered to be to highly mineralized for human or livestock consumption.

The best quality water available to Marion County residents, drawn from bedrock sources, is from the Mississippian aquifer where it is not overlain by Pennsylvanian rocks, and from the Cambro-Ordovician aquifer. However, there may be local exceptions with respect to these.

Ground water from rock aquifers in Marion County is hard owing to either high iron or Calcium carbonate content. High fluoride concentration's may be a problem with water from the lower part of the Mississippian in certain areas and from the Cambro-Ordovician in the southwestern part of the county. The temperature of ground water from the Mississippian ranges from 52°F to 64°F (11°C to 18°C) and averages about 57°F (14°C). The temperature for water from the Cambro-Ordovician aquifer is higher, ranging between 73°F and 76°F (23°C and 24°C).

RECOMMENDATIONS FOR PRIVATE WATER WELLS

Contracting for Well Construction

To protect your investment and guarantee satisfactory well completion, it is a good idea to have a written agreement with the well driller. The agreement should specify in detail:

size of well, casing specifications, and types of screen and well seal

methods of eliminating surface and subsurface contamination

disinfection procedures to be used

type of well development if necessary

test pumping procedure to be used

date for completion

itemized cost list including charges for drilling per foot, for materials per unit, and for other operations such as developing and test pumping

guarantee of materials, workmanship, and that all work will comply with current recommended methods

liability insurance for owner and driller

Well Location

A well should be located where it will be least subject to contamination from nearby sources of pollution. The Iowa State Department of Health recommends minimum distances between a new well and pollution sources, such as cesspools (150 ft.), septic tanks (50 ft.), and barnyards (50-100 ft. and downslope from well). Greater distances should be provided where possible.

The well location should not be subject to flooding or surface water contamination. Select a well-drained site, extend the well casing a few feet above the ground, and mound earth around it. Diversion terraces or ditches may be necessary on slopes above a well to divert surface runoff around the well site.

In the construction of all wells, care should be taken to seal or grout the area between the well bore and the well casing (the annulus), as appropriate, so that surface water and other pollutants cannot seep into the well and contaminate the aquifer.

Locate a well where it will be accessible for maintenance, inspection, and repairs. If a pump house is located some distance from major buildings and wired separately for power, continued use of the water supply will not be jeopardized by fire in major buildings.

Water Treatment

Water taken from a private well should ideally be tested every six months. The University Hygienic Laboratory will do tests for coliform bacteria, nitrate, iron, hardness, and iron bacteria in drinking water for private individuals. Special bottles must be used for collecting and sending water samples to the laboratory. A sample kit can be obtained by writing to the University Hygienic Laboratory, University of Iowa, Oakdale Campus, Iowa City, Iowa 52242. Indicate whether your water has been treated with chlorine, iodine, or bromine; for different sample bottles must be used for treated and untreated water. The charge for the bacterial test is \$3.00; for iron hardness and nitrate, it is \$3.00; and for iron bacteria, \$5.00. If your well is determined to be unsafe, advice for correcting the problem can be obtained from your county or state Department of Health. Several certified private laboratories also run water analyses.

Shock chlorination is recommended following the construction and installation of a well and distribution system and anytime these are opened for repairs or remodeling a strong chlorine solution is placed in the well and complete distribution system to kill nuisance and disease-causing organisms. If the first shock chlorination does not rid the water supply of bacteria it should be repeated, if this does not solve the problem the well should be abandoned or the water should be continuously disinfected with proper chlorination equipment.

Since most of the ground waters in Marion County are mineralized, water softening and iron removal equipment may make water more palatable and pleasant to use. Softened water contains increased sodium; contact your physician before using a softener if you are on a sodium-restricted diet. Chlorination followed by filtration will remove most forms of iron and iron bacteria. Iron bacteria has no adverse effect on health but will plug wells, water lines, and equipment and cause tastes and odors. Iron removal equipment can be used if problems persist.

Well Abandonment

Wells taken out of service provide easy access for pollution to enter aquifers supplying water to other wells in the vicinity. Unprotected wells may also cause personal injury. Proper abandonment procedures should be followed to restore the natural conditions that existed before well construction and prevent any future contamination. Permanent abandonment requires careful sealing. The well should be filled with concrete, cement grout, or sealing clays throughout its entire length. Before dug or bored wells are filled at least the top 10 feet of lining should be removed so surface waters will not penetrate the subsurface through a porous lining or follow cracks in or around the lining. The site should be completely filled and mounded with compacted earth.

ABANDONED WELLS SHOULD NEVER BE USED FOR DISPOSAL OR SEWAGE OR OTHER WASTES.

SOURCES OF INFORMATION

In planning the development of a ground water supply or contracting for the drilling of a new well additional or more specific information is often required. This report section lists several sources and types of additional information.

State Agencies That May Be Consulted

Iowa Geological Survey ¹	123 North Capitol Iowa City 52242	(319) 338-1173
State Health Department ^{2,6}	Lucas Building Des Moines 50319	(515) 281-5787
Iowa Natural Resources Council ³	Wallace Building Des Moines 50319	(515) 281-5914
Iowa Dept. of Environ. Quality ⁴	Wallace Building Des Moines 50319	(515) 281-8854
University Hygienic Laboratory ⁵	U. OF IA, Oakdale Campus Iowa City 52242	(319) 353-5990
Cooperative Extension Service in ⁶	110 Curtis Hall, ISU Ames 50011	(515) 294-4569

Functions:

- 1 Geologic and ground water data repository, consultant on well problems, water development and related services
- 2 Drinking water quality, public and private water supplies
- 3 Water withdrawal regulation and Water Permits for wells withdrawing more than 5000 gpd
- 4 Municipal supply regulation and well construction permits
- 5 Water quality analysis
- 6 Advice on water systems design and maintenance

Well Drillers and Contractors

The listing provided here was drawn from an Iowa Geological Survey mailing list and yellow pages of major towns in phone books. These selected are within an approximate radius of 50 miles of Marion County. For a state-wide listing contact either the Iowa Water Well Drillers Association, 4350 Hopewell Ave., Bettendorf, Iowa 51712, (319) 355-7528 or the Iowa Geological Survey, (319) 338-1173.

Ahrens Well Drilling
R.R.# 2
Montezuma, IA 50171

Brooks Well and Pump Co.
Knoxville, IA 50138

Douglas Bruinekool
Bruinekool Well Co.
Pella, IA 50219

Dwayne Bruinekool
Bruinekool Well Co.
Oskaloosa, IA 52577

Hughes Well Co.
4120 73rd
Des Moines, IA 50322

Moorehead Well Co.
R.R. #1
Indianola, IA 50125

Whalen Well Co.
1407 1st Ave.
Newton, IA 50208

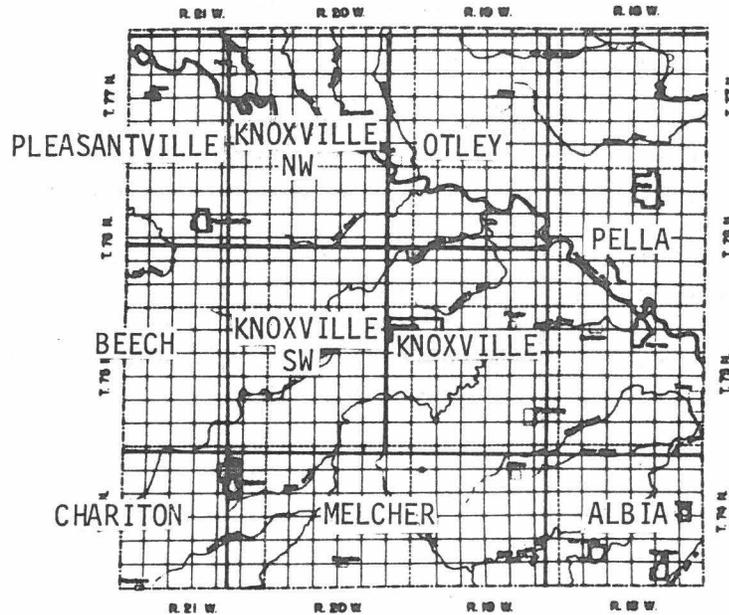
Snook Well Co.
R.R. #1
Promise City, IA 52583

Thorpe Well Co.
Ankeny, IA 50021

Doyle Van De Krol
Sully, IA 50251

Verwers Well Co.
Sully, IA 50251

Topographic Maps (Available from the Iowa Geological Survey)



<u>Map Title</u>	<u>Date</u> (Published)	<u>Scale</u>	<u>Contour Interval</u>
Albia	1926	1:62,500	20'
Chariton	1915	1:62,500	20'
Melcher	1916-22	1:62,500	20'
Pella	1910	1:62,500	20'
Beech	1965	1:24,000	10'
Knoxville	1965	1:24,000	10'
Knoxville SW	1965	1:24,000	10'
Knoxville NW	1965	1:24,000	10'
Otley	1965	1:24,000	10'
Pleasantville	1965	1:24,000	10'

Useful Reference Materials

Cagle, Joseph W., and Heintze, A.J., 1978, Water Resources of South-central Iowa, Iowa Geological Survey, Water Atlas No. 5.

Horick, P.J., and Steinhilber, W.L., 1973, Mississippian aquifer of Iowa, Iowa Geological Survey, Misc. Map Series No. 3.

Horick, P.J., and Steinhilber, W.L., 1978, Jordan aquifer of Iowa, Iowa Geological Survey, Misc. Map Series No. 6.

Iowa State Department of Health, 1971 Sanitary standards for water wells, State Department of Health, Environmental Engineering Service.

Van Eck, O.J., 1971, Optimal well plugging procedures, Iowa Geological Survey, Public Information Circular No. 1.

Van Eck, O.J., 1978, Plugging procedures for domestic wells, Iowa Geological Survey, Public Information Circular No. 11.