GROUND WATER RESOURCES

Henry County

Open File Report 80-44 WRD

Compiled by PATRICIA M. WITINOK

GROUND-WATER RESOURCES OF HENRY COUNTY

Introduction

Approximately 99% of the residents of Henry County rely on ground water as the source of their drinking water. It is estimated that the use of ground water in the county currently approaches .96 billion gallons per year. For comparison, this amount would provide each resident with 142 gallons of water a day during the year. Actually, few if any households use this much water, and the rather large annual per capita use reflects the greater water requirements of the county's industries, agribusinesses, and municipalities.

The users of ground water in the county draw their supplies from several different geologic sources. Several factors must be considered in determining the availability of ground water and the adequacy of a supply source:

distribution - having water where it is needed,

accessibility - affects the costs for drilling wells and pumping water,

yield - relates to the magnitude of the supply that can be sustained,

quality - determines for what purposes the water can be used.

In terms of these factors, there are few locations in Henry County where the availability of ground water is not limited to some degree. The most common limitation is poor water quality, that is, highly mineralized ground water. Secondary limitations are generally related to poor distribution, small yeilds from some sources, and poor accessibility due to the great depths to adequate sources.

Occurrence of Ground Water in Henry County

The occurrence of ground water is influenced by geology -- the position and thickness of the rock units, their ability to store and transmit water, and their physical and chemical make-up. Geologic units that store and transmit water and yield appreciable amounts to wells are called aquifers. The best aquifers are usually composed of unconsolidated sand and gravel, porous sandstone, and porous or fractured limestone and dolostone. Other units with materials such as clay and silt, shale, siltstone, and mudstone yield little or no water to wells. These impermeable units are called aquicludes or aquitards and commonly separate one aquifer unit from another.

In Henry County there are four principal aquifers from which users obtain water supplies. The loose, unconsolidated materials near the land surface comprise the surficial aquifer. Below this there are three major rock aquifers -- the Mississippian, the Devonian, and the Cambro-Ordovician aquifers. Figure 1 shows the geologic relations of these beneath the county. Each of the aquifers has its own set of geologic, hydrologic, and water quality characteristics which determine the amount and potability (suitability for drinking) of water it will yield.

Surficial Aquifers

Unconsolidated deposits at the land surface are comprised of mixtures of clay, silt, sand, gravel, and assorted boulders. Water-yielding potential of the surficial deposits is greatest in units composed mostly of sand and/or gravel. Three types of surficial aquifers are used: the alluvial aquifer, the drift aquifer, and the buried channel aquifer.

The alluvial aquifer consists mainly of the sand and gravel transported and deposited by modern streams and makes up the floodplains and terraces in major valleys. Alluvial deposits are shallow, generally less than 50-60 feet and thus may be easily contaminated by the infiltration of surface water.

The drift aquifer is the thick layer of clay to boulder size material deposited over the bedrock by glacial ice which invaded the county at least twice in the last two million years. The composition of the glacial drift varies considerably and in many places does not yield much water. There are, however, lenses or beds of sand and gravel within the drift which are thick and widespread enough to serve as dependable water sources. These lenses are difficult to locate because they are irregular in shape and buried within the drift deposits. Usually one or two sand layers can be found in most places that will yield minimum water supplies for domestic wells.

The buried channel aquifer consists of stream alluvium of partially filled valleys that existed before the glacial period. The valleys were overridden by the glaciers and are now buried under glacial and recent alluvial deposits.

The distribution, yields, and water quality characteristics for the surficial aquifers are summarized in Figures 2, 9, and 13. An indication of accessibility can be obtained by comparing the elevations of the top (the land surface) and the bottom (the bedrock surface) of the surficial deposits from Figures 4 and 5. The thickness of the glacial drift and the depth of the buried channels are determined by subtracting the elevations at selected locations.

Rock Aquifers

Below the surficial materials is a thick sequence of layered rocks formed from deposits of rivers and shallow seas that have covered the state within the last 600 million years. The geologic map (Figure 3) shows the geologic units which form the top of this rock sequence. These rocks are Pennsylvanian in age and are mainly shales. Although the Pennsylvanian rocks usually act as an aquiclude, there are sparse local sandstone layers (in the northwest at Wayland, approximately 8' thick and moving just south of there at Rome, approximately 15' thick) which could supply small yields to domestic wells. The Pennsylvanian rocks are very patchy throughout the county, concentrated mostly in the southwestern corner at about its thickest point of 40' and dwindling to 0' in most other places, except for a small patch in the extreme northwest part of the county.

Underlying the Pennsylvanian aquiclude is a sequence of older rocks, portions of which form the three major rock aquifers in Henry County. This sequence and its water-bearing characteristics are shown in Table 1.

Examples of the rock units encountered in several wells at various locations in Henry County are indexed and illustrated in Figures 7 and 8. The geologic unit that supplies ground water and the rate of yield are shown for each well.

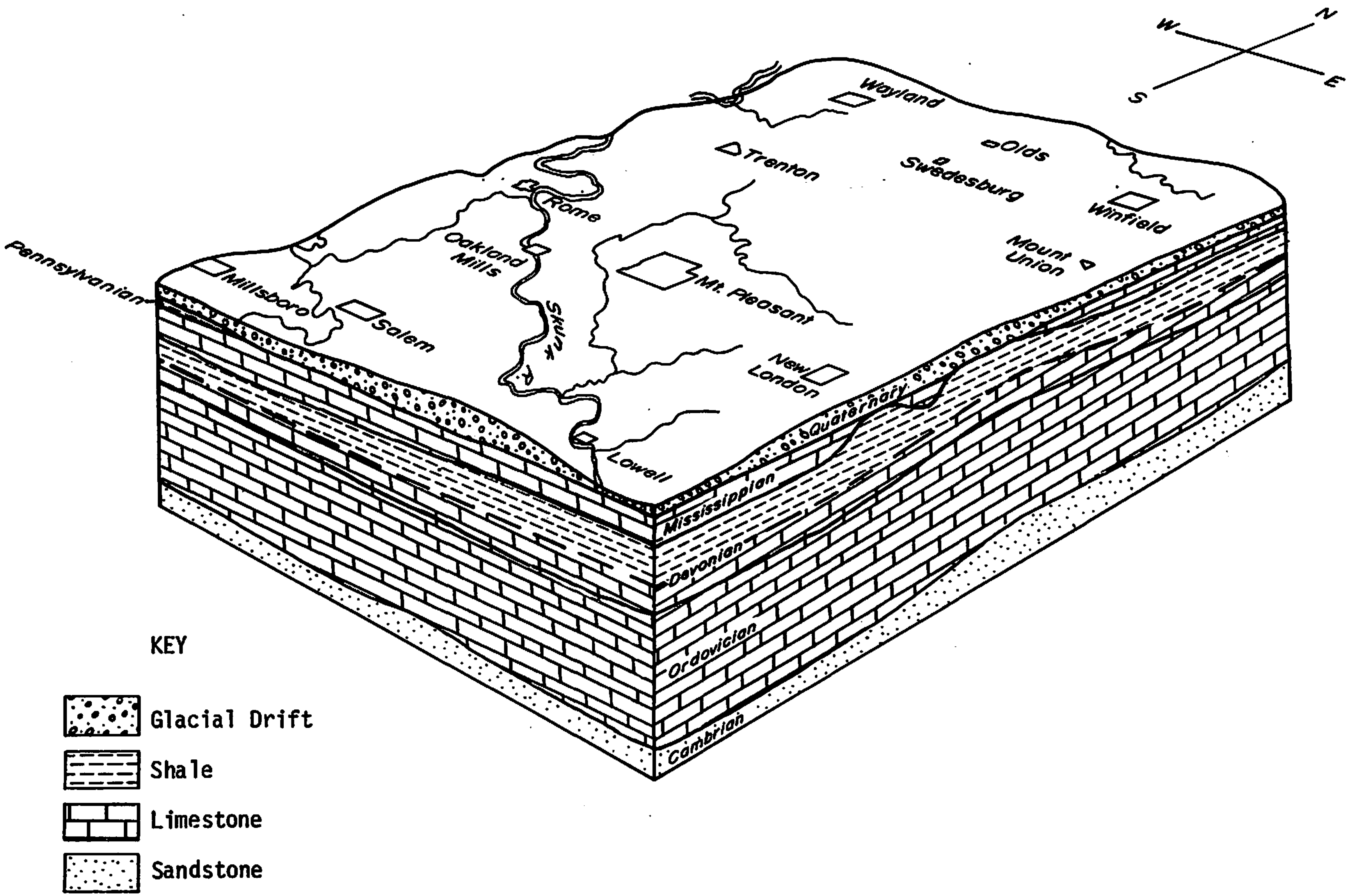


Figure 1

BLOCK DIAGRAM SHOWING THE GEOLOGY OF HENRY COUNTY

The accessibility of ground water in rock aquifers depends on the depth to the aquifer. The deeper a well must be, the greater the cost for well construction and pumping. The depths to and thicknesses of units at specific sites will vary somewhat because of irregularities in the elevation of the land surface and in the elevation of the tops of the underlying rock units. Estimates of depths and thicknesses can be made by comparing Figure 4 with the maps of aquifer elevations in Figures 10, 11 and 12. The range in depth below land surface to the top of the county's principal bedrock aquifers is given for each township in Figure 6.

A second factor which affects ground water accessibility is the level to which the water will rise in a well (the static water level). Throughout the county water in the rock aquifers is under artesian pressure and rises in the well once the aquifer is penetrated. This can reduce the cost of pumping. Average static water levels for Henry County wells are shown in Figures 10, 11 and 12.

Average rates of yield and water quality characteristics for each of the aquifers are summarized in the maps in Figures 10, 11, 12, 13, 14 and 15.

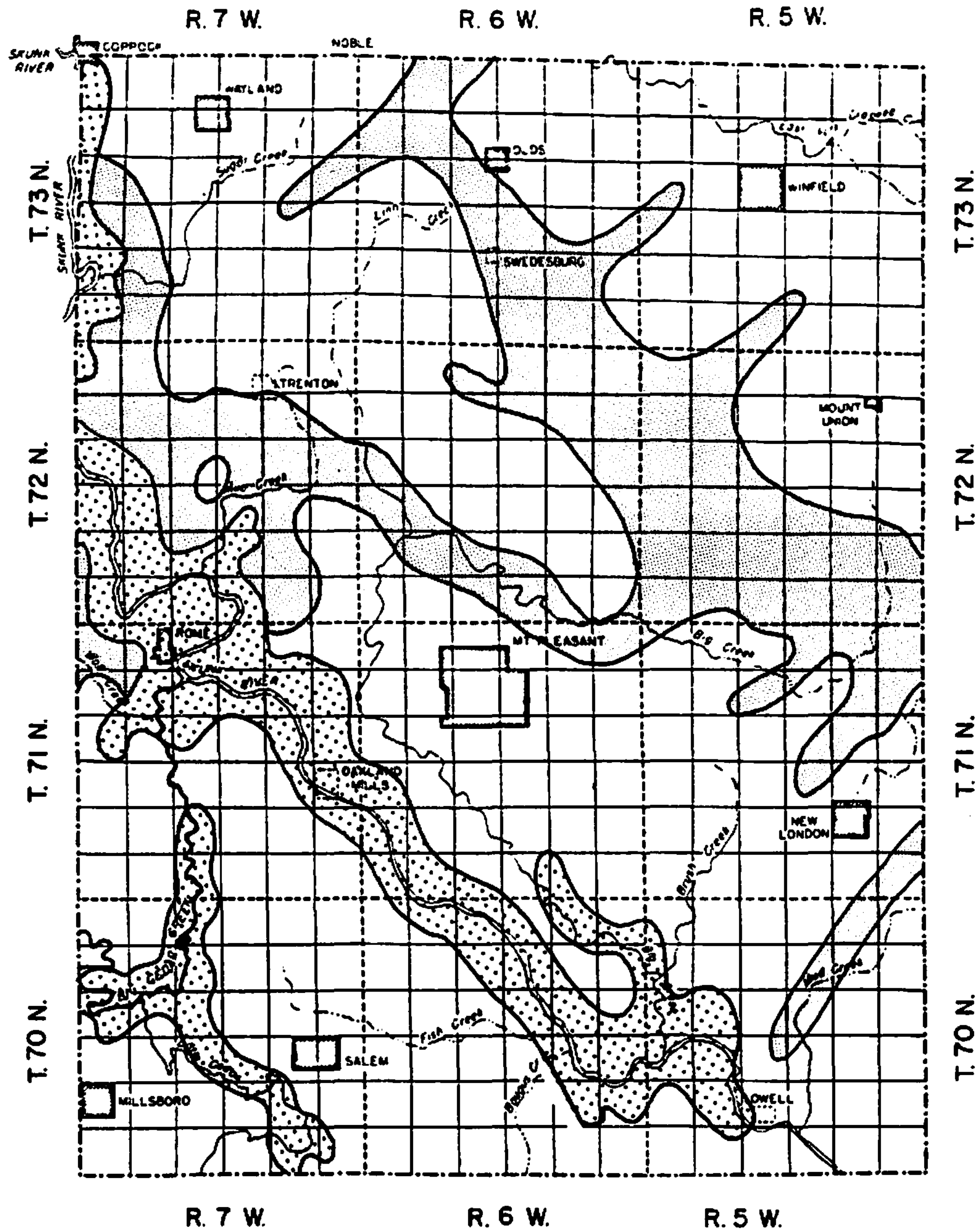
Table 1

GEOLOGIC AND HYDROGEOLOGIC UNITS IN HENRY COUNTY

Age	Rock Unit	Description	Thickness Range	Hydrogeologic Unit	Water-Bearing Characteristics
Quaternary	Alluvium	Sand, gravel, silt and clay	0-250 (feet)	Surficial aquifer	Fair to large yields (25 to 100 gpm)
	Glacial drift (undifferentiated)	Predominantly till containing scattered irregular bodies of sand and gravel			Low yields (less than 10 gpm)
	Buried channel deposits	Sand, gravel, silt and clay			Small to large yields
Pennsylvanian	Des Moines Series	Shale, sandstones; mostly thin and patchy	0-40	Aquiclude	Low yields only from limestone and sandstone
Mississippian	Meramec Series	Sandy limestone	0-300	Mississippian aquifer	Fair to low yields
	Osage Series	Limestone and dolostone cherty; shale			
	Kinderhook Series	Limestone, oolitic and dolostone, cherty, also siltstone			
Devonian	Maple Mill Shale Sheffield Formation Lime Creek Formation	Shale; limestone in lower part	280-325	Devonian aquiclude	Does not yield water
	Cedar Valley Limestone Wapsipinicon Formation	Limestone and dolostone; contains evaporites (gypsum) in southern half of Iowa	125-275	Devonian aquifer	Fair to low yields
Ordovician	Maquoketa Formation	Shale and dolostone	875-1075	Maquoketa aquiclude	Does not yield water
	Galena Formation	Dolostone and chert		Minor aquifer	Low yields
	Decorah Formation-Platteville Formation	Limestone, dolostone and thin shale, includes sandstone in SE Iowa		Aquiclude	Does not yield water
	St. Peter Sandstone	Sandstone		Cambrian-Ordovician aquifer	Fair yields
	Prairie du Chien Formation	Dolostone, sandy and cherty			High yields (over 500 gpm)
Cambrian	Jordan Sandstone	Sandstone	60-120		
	St. Lawrence Formation	Dolostone		Aquitard	Low yields
	Franconia Sandstone	Sandstone and shale			
	Dresbach Group	Sandstone		Dresbach aquifer	High to low yields
Precambrian	Undifferentiated	Coarse sandstones; crystalline rocks		Base of groundwater reservoir	Not known to yield water

Figure 2

SURFICIAL MATERIALS



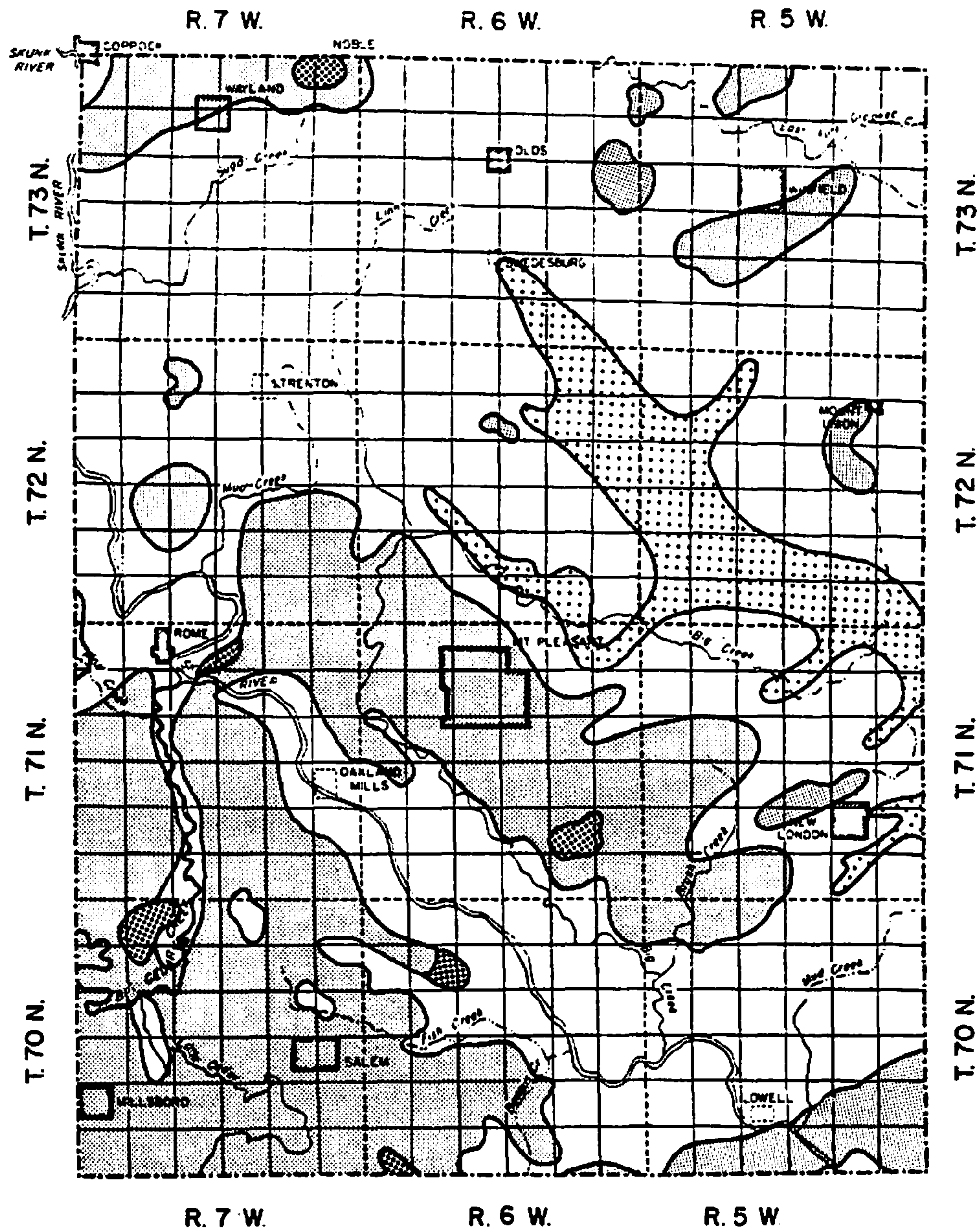
Alluvium

Glacial Drift

Buried Channels

Figure 3

GEOLOGIC MAP



Pennsylvanian

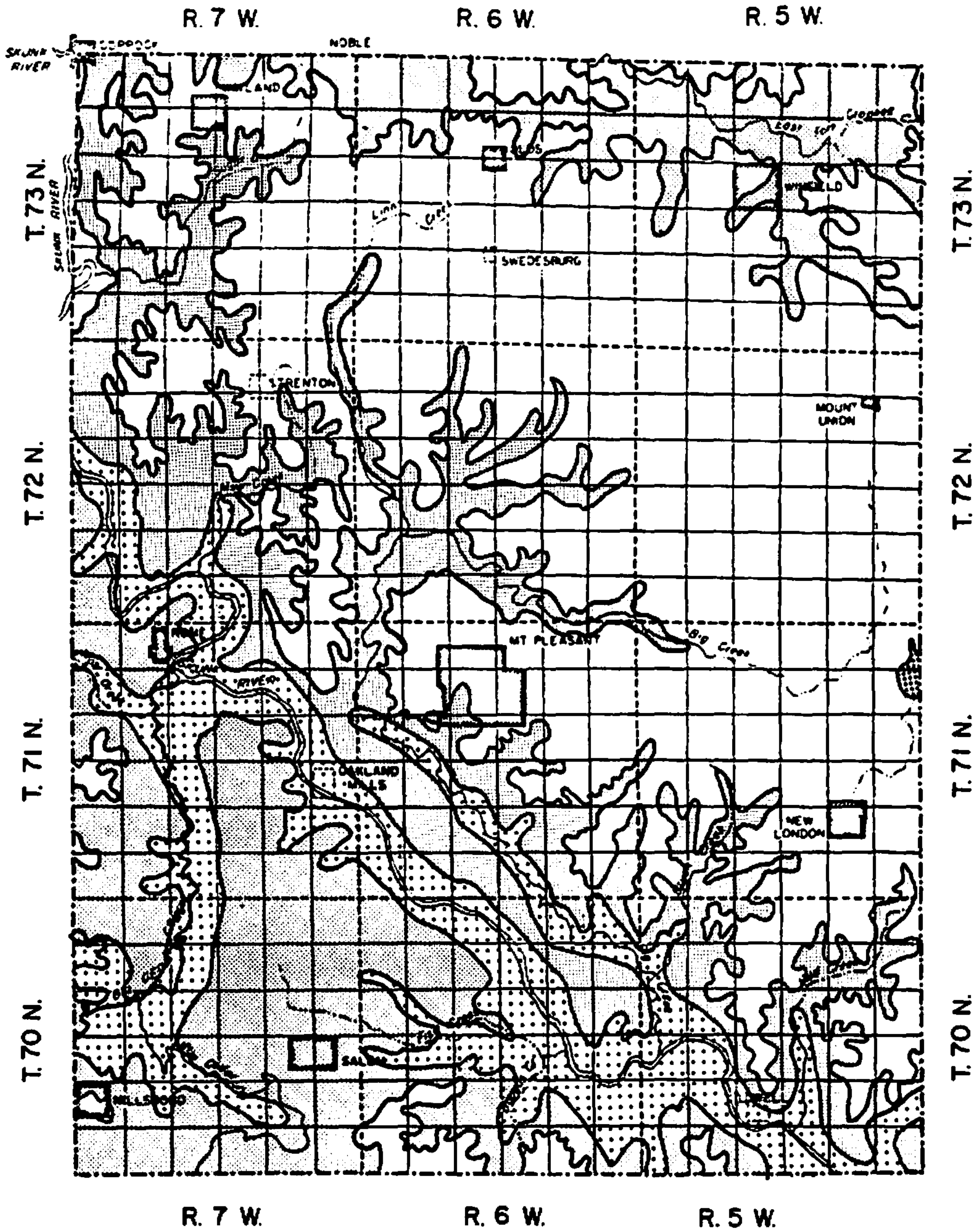
Mississippian - Lower Part

Mississippian - Upper Part

Devonian - Mississippian Aquiclude

Figure 4

ELEVATION OF LAND SURFACE IN FEET ABOVE MEAN SEA LEVEL



500-600

700-800

600-700

800-900

Figure 5

ELEVATION OF BEDROCK SURFACE IN FEET ABOVE MEAN SEA LEVEL

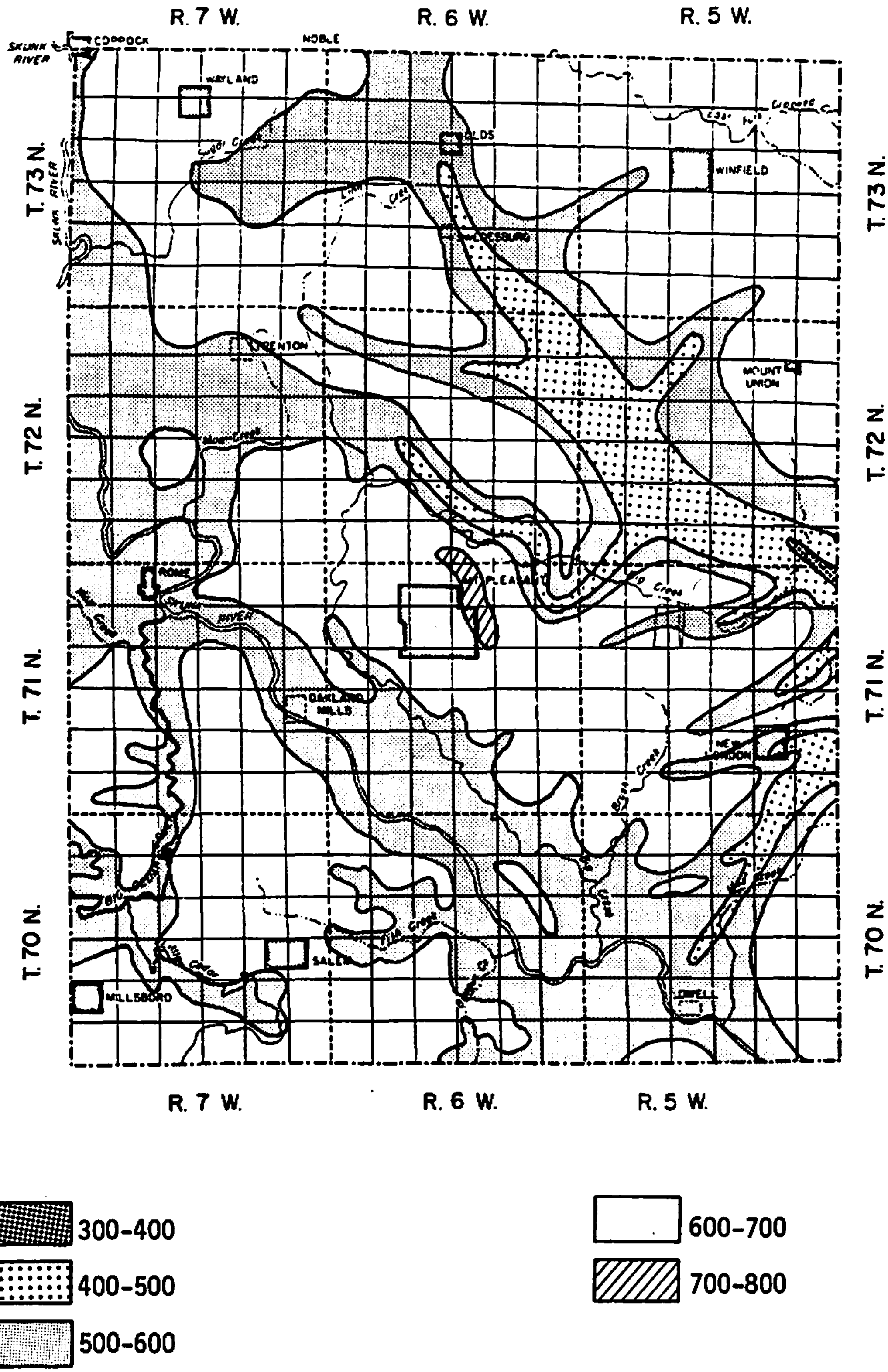


Figure 6

RANGE IN DEPTH TO HENRY COUNTY'S PRINCIPAL ROCK AQUIFERS

	R.7W.	R.6W.	R.5W.	
T.73N.	BEDROCK 0-200 MISSISSIPPIAN 50-200 DEVONIAN 300-600 CAMBRO-ORDOVICIAN 1000-1300	BEDROCK 0-300 MISSISSIPPIAN 50-200 DEVONIAN 300-600 CAMBRO-ORDOVICIAN 1050-1250	BEDROCK 0-200 MISSISSIPPIAN 25-200 DEVONIAN 200-600 CAMBRO-ORDOVICIAN 1000-1200	T.73N.
T.72N.	BEDROCK 0-200 MISSISSIPPIAN 50-200 DEVONIAN 300-700 CAMBRO-ORDOVICIAN 1000-1300	BEDROCK 0-300 MISSISSIPPIAN 25-300 DEVONIAN 200-700 CAMBRO-ORDOVICIAN 1000-1200	BEDROCK 0-300 MISSISSIPPIAN 50-300 DEVONIAN 200-600 CAMBRO-ORDOVICIAN 1000-1200	T.72N.
T.71N.	BEDROCK 0-100 MISSISSIPPIAN 0-200 DEVONIAN 250-600 CAMBRO-ORDOVICIAN 1000-1300	BEDROCK 0-200 MISSISSIPPIAN 0-200 DEVONIAN 200-600 CAMBRO-ORDOVICIAN 1000-1200	BEDROCK 0-300 MISSISSIPPIAN 50-300 DEVONIAN 300-700 CAMBRO-ORDOVICIAN 1000-1300	T.71N.
T.70N.	BEDROCK 0-100 MISSISSIPPIAN 50-150 DEVONIAN 300-700 CAMBRO-ORDOVICIAN 1000-1300	BEDROCK 0-200 MISSISSIPPIAN 0-200 DEVONIAN 300-700 CAMBRO-ORDOVICIAN 1000-1300	BEDROCK 0-300 MISSISSIPPIAN 50-300 DEVONIAN 300-700 CAMBRO-ORDOVICIAN 900-1200	T.70N.
	R.7W.	R.6W.	R.5W.	

Figure 7

INDEX MAP FOR TYPICAL WELLS IN HENRY COUNTY

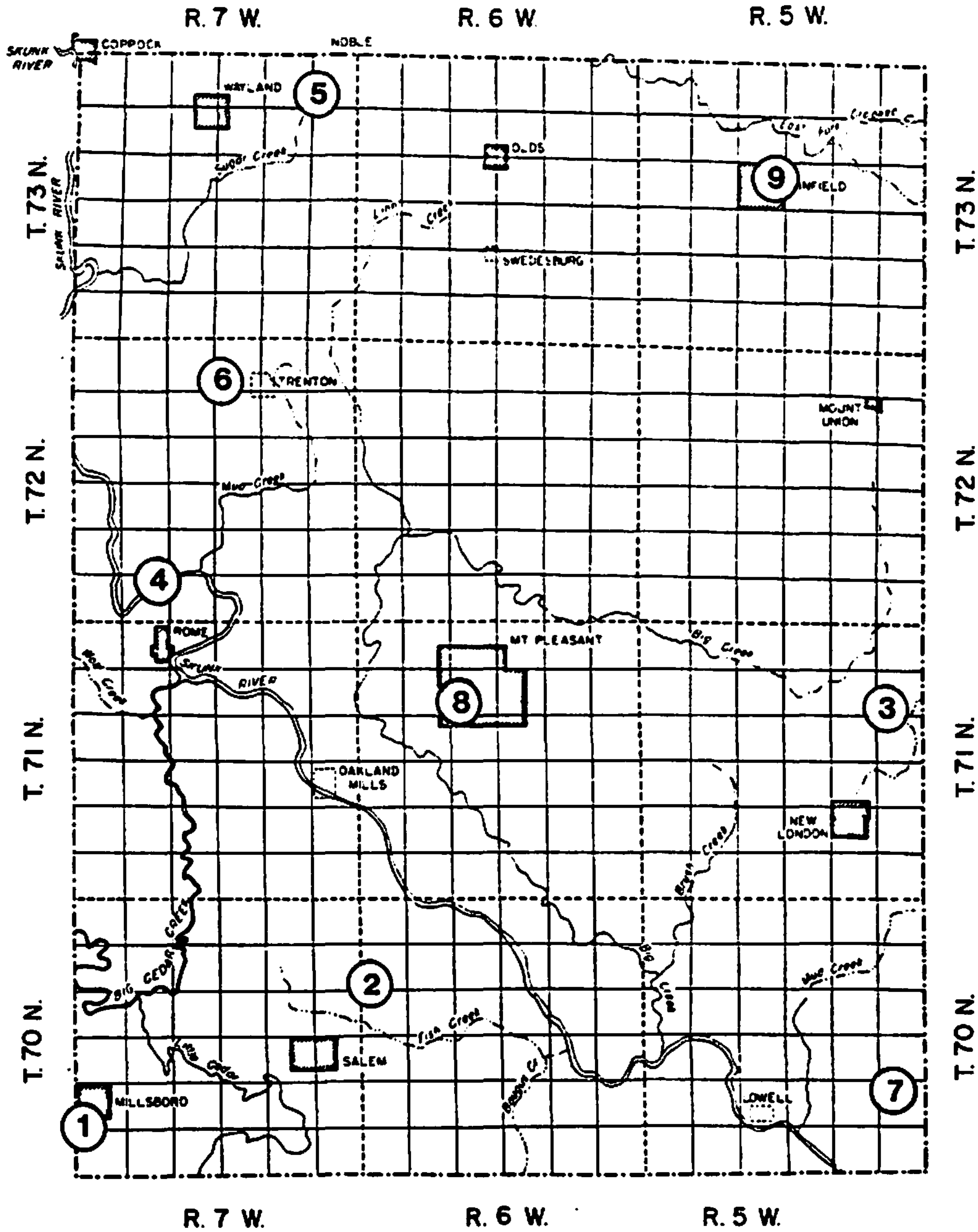


Figure 8

TYPICAL WELLS IN HENRY COUNTY

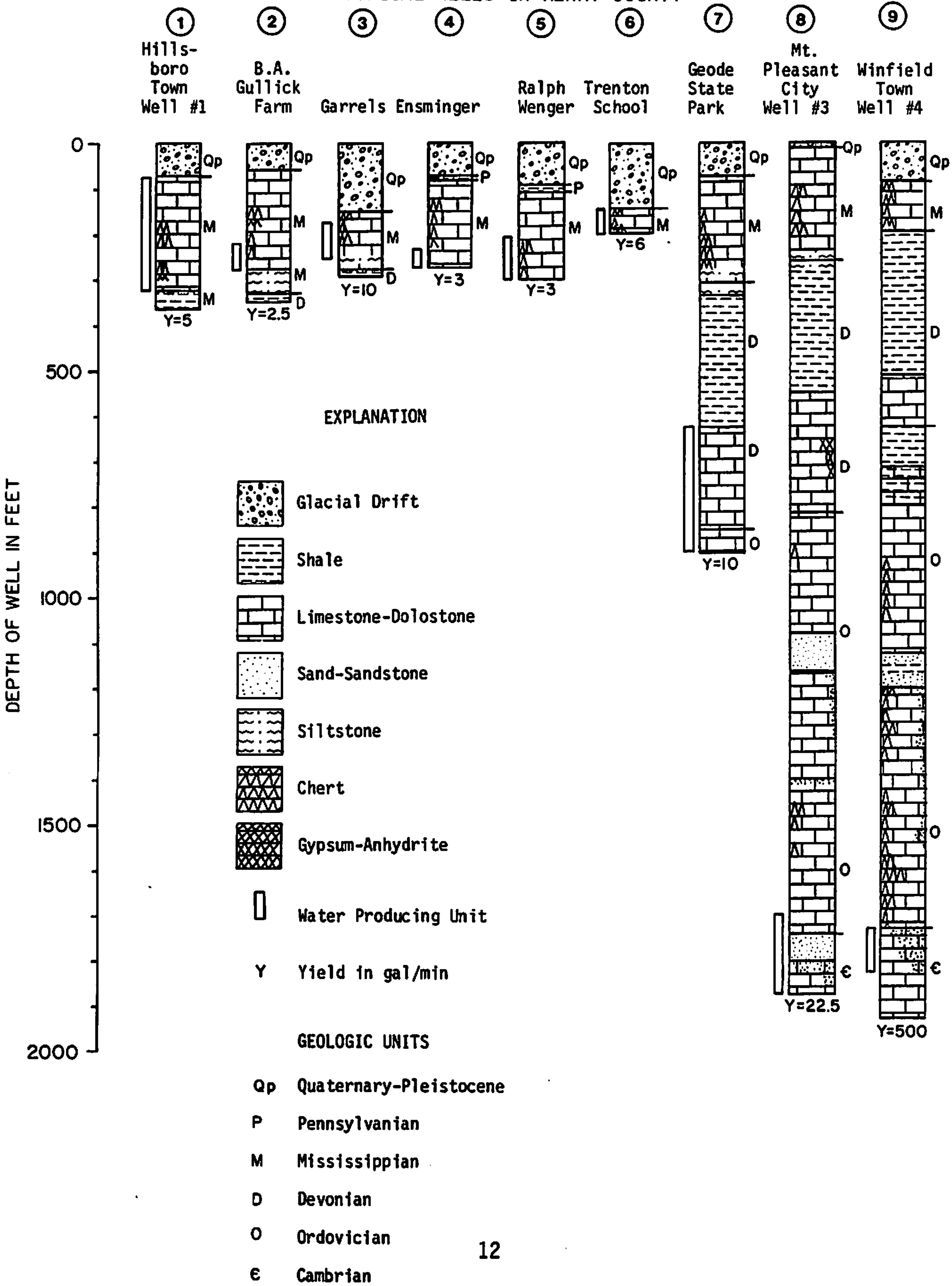


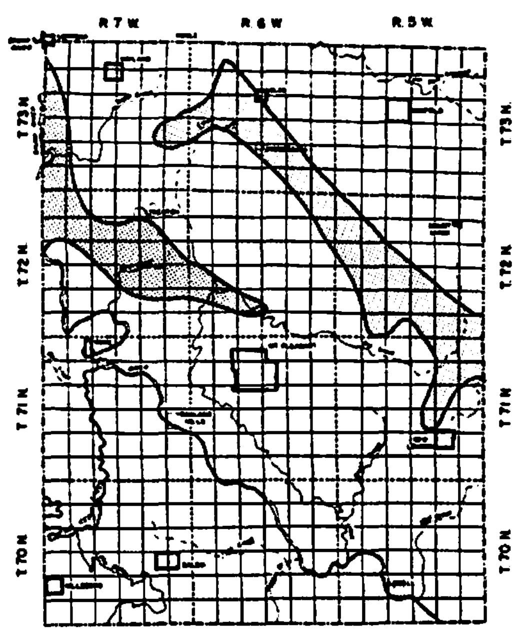
Figure 9

SURFICIAL AQUIFERS

Water Levels

Water levels in the surficial aquifers are difficult to analyze, water rises to different levels in wells drilled into alluvial, buried-channel, and drift aquifers. The water table in the drift aquifer generally slopes from high land areas toward the streams and, changes noticeably throughout the year. Levels in drift and buried-channel aquifers respond rapidly to recharge from precipitation. Water levels in the alluvial aquifer fluctuates somewhat in the same way as those in the drift and buried-channel aquifers; however, the main influence on the alluvial aquifer is the stage (level) of the associated streams. Water levels will be high during periods of high stream stage and low during the low-stage periods.

Water levels in the drift aquifers commonly are from 10 to 50 feet below the land surface, and those in the buried-channel aquifers have been reported to be as low as 175 feet below the flood-plain surface and the depth to the water surface will be accordingly deeper in wells located on terrace surfaces.



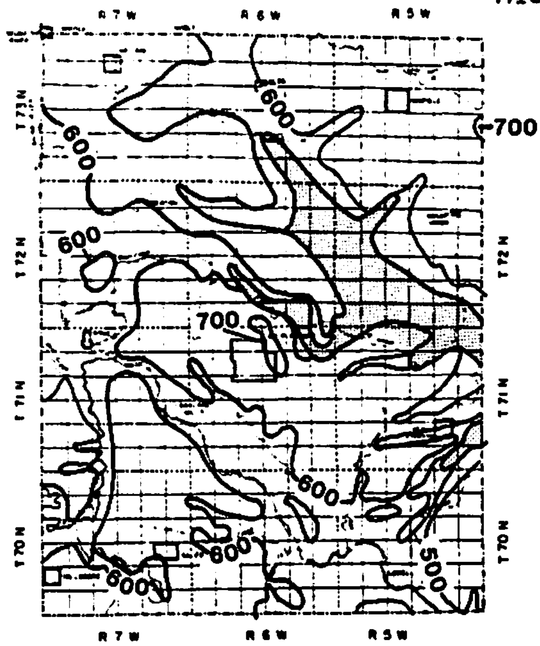
Water yields to wells in gallons per minute

Below 20

20-100

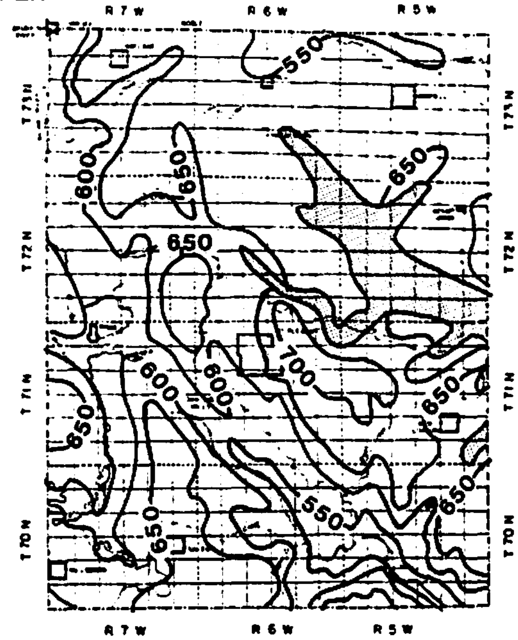
Figure 10

MISSISSIPPIAN AQUIFER



Elevation of Mississippian Aquifer in feet above mean sea level

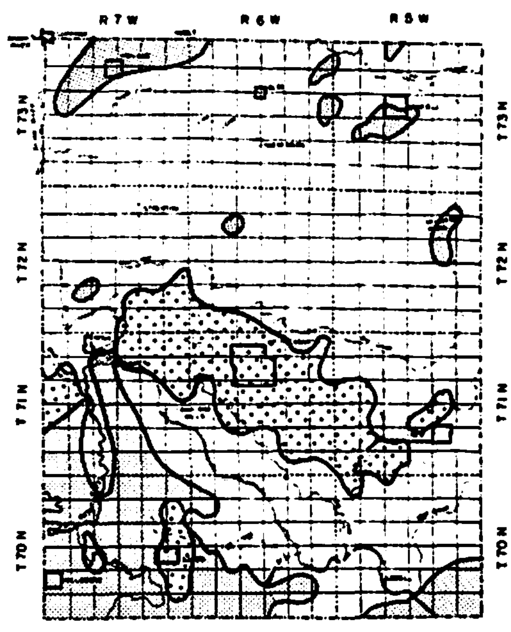
Aquifer not present



Water levels in wells in feet above mean sea level

Aquifer not present

Upper Part

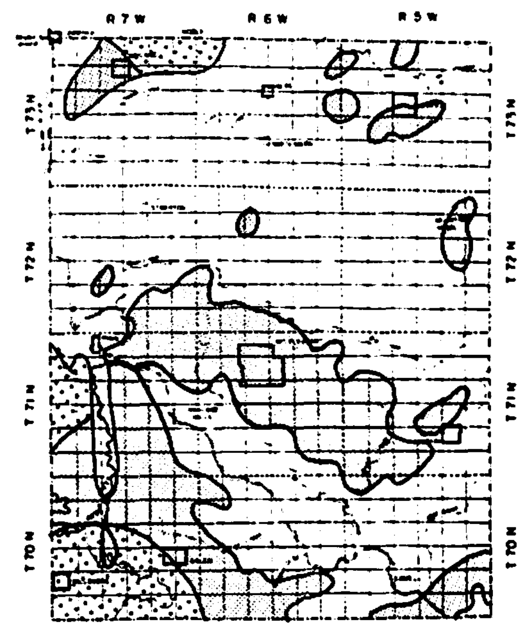


Water yields to wells in gallons per minute

Below 20

20 - 50

Aquifer not present



Dissolved solids content in milligrams per liter (mg/l)*

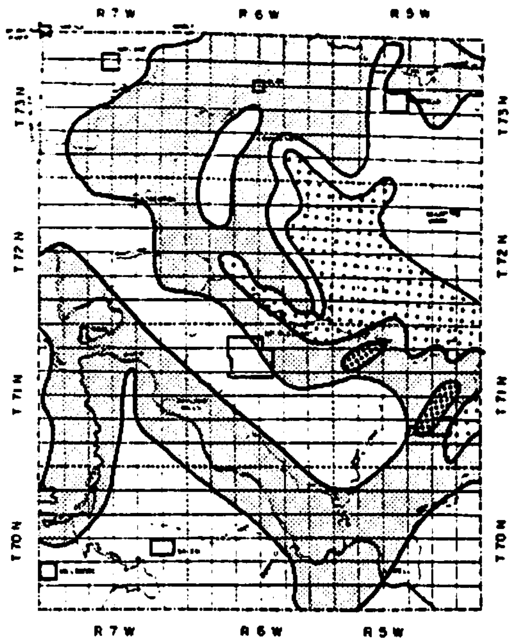
Below 500

500 - 1000

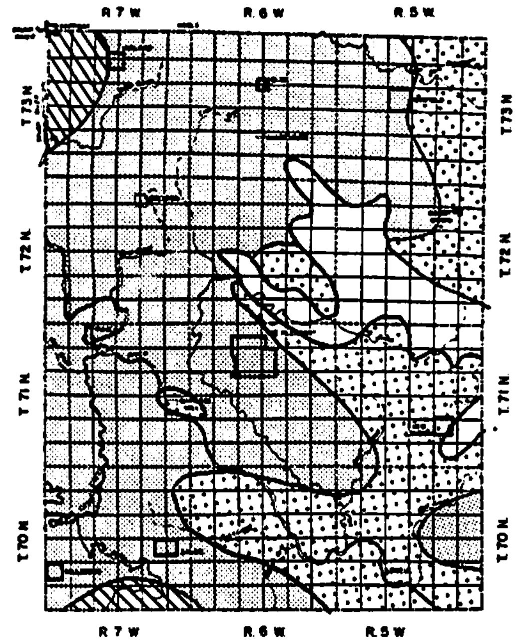
Aquifer not present

*Other water quality data in Figure 13

MISSISSIPPIAN AQUIFER



Lower Part



Water yields to wells in gallons per minute

- Below 20
- 20 - 50
- more than 50
- Aquifer not present

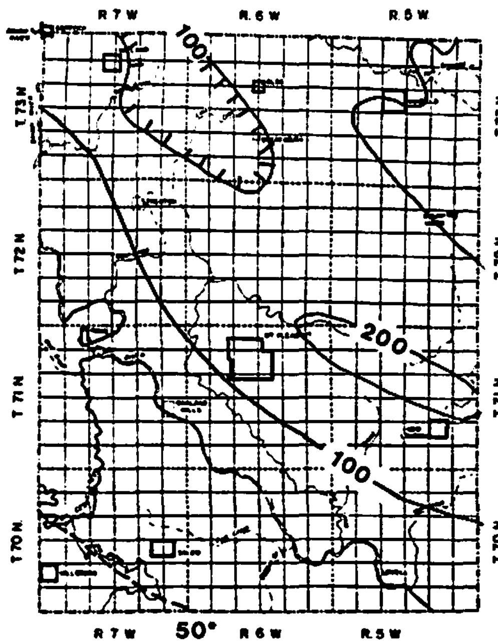
Dissolved solids content in milligrams per liter (mg/l)*

- Below 500
- 500 - 1000
- 1000 - 2500
- Aquifer not present

*Other water quality data in Figure 14.

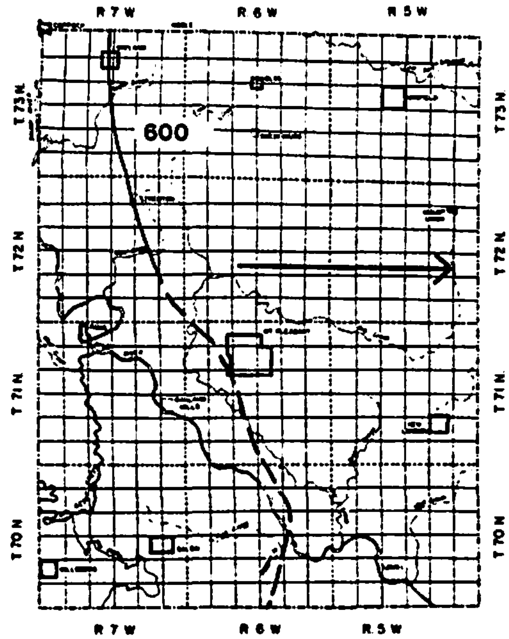
Figure 11

DEVONIAN AQUIFER



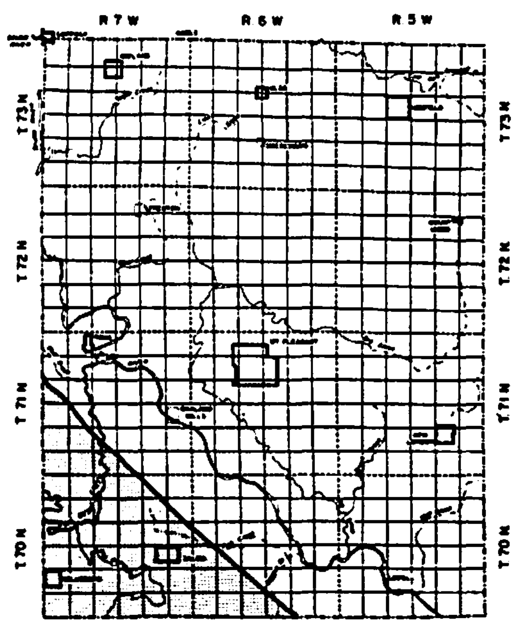
Elevation of the Devonian Aquifer in feet above mean sea level

----- 50' contour line showing decrease in elevation of aquifer to SW.



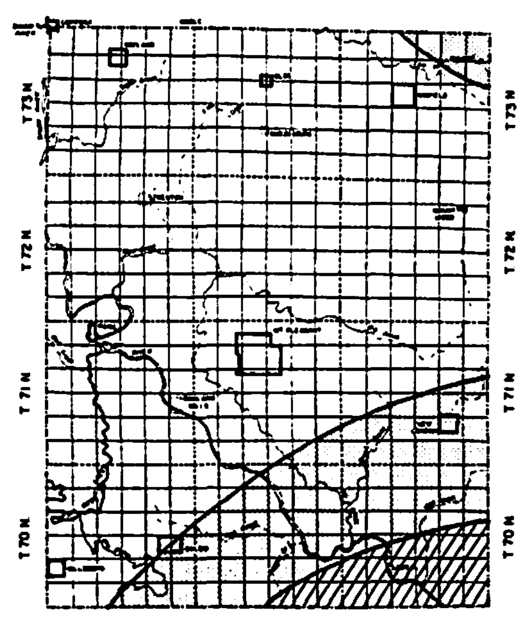
Water levels in wells in feet above mean sea level

⟶ Arrow showing direction of decrease in piezometric surface, by 50' contour interval



Water yields to wells in gallons per minute

- Below 20
- No Data Available



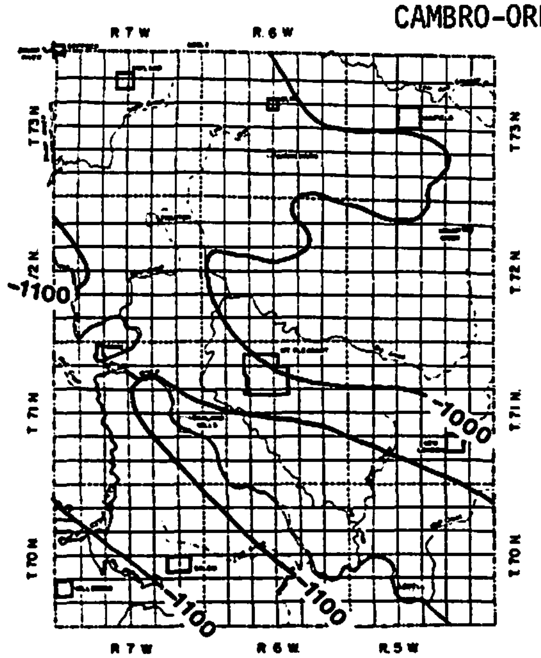
Dissolved solids content in milligrams per liter (mg/l)*

- 2500 - 5000
- 5000 - 7500
- 7500 - 10,000

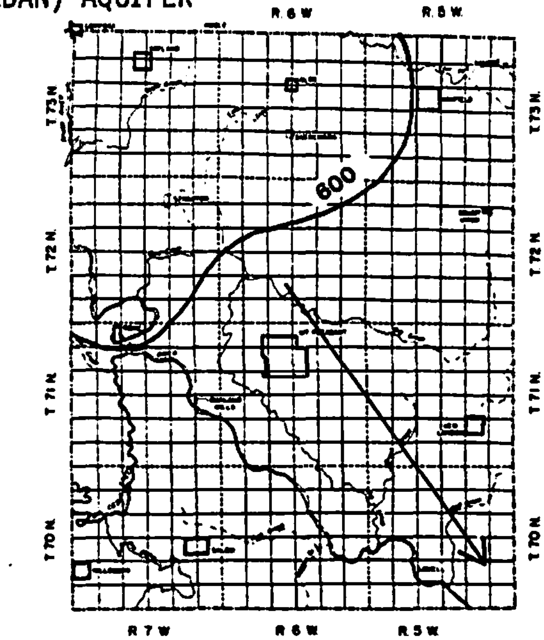
*Other water quality data in Figure 15

Figure 12

CAMBRO-ORDOVICIAN (JORDAN) AQUIFER

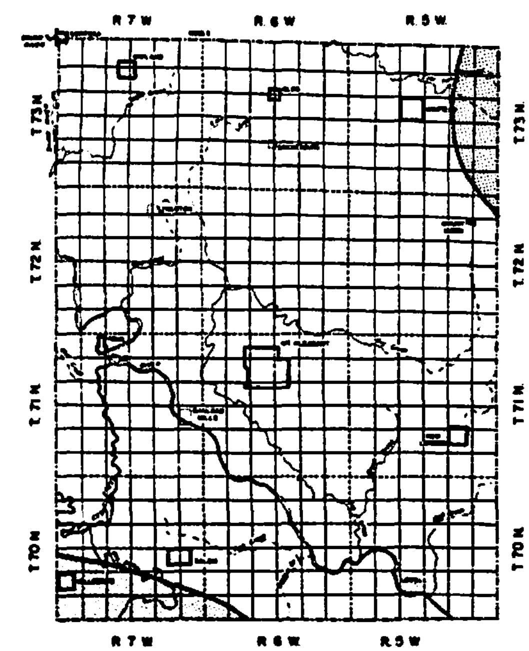


Elevation of the Jordon aquifer in feet above mean sea level



Water levels in wells in feet above mean sea level

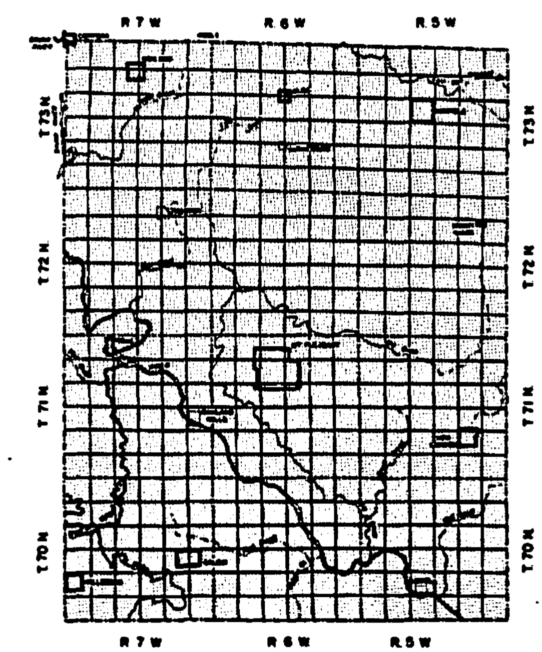
⟶ Arrow showing direction of decrease in piezometric surface, by 50' contour interval



Water yields to wells in gallons per minute

500 - 1000

More than 1000



Dissolved solids content in milligrams per liter (mg/l)*

1000 - 1500

*Other water quality data in Figure 15

Table 2

SIGNIFICANCE OF MINERAL CONSTITUENTS AND PHYSICAL PROPERTIES OF WATER

Constituent or Property	Maximum Recommended Concentration	Significance
Iron (Fe)	0.3 mg/l	Objectionable as it causes red and brown staining of clothing and porcelain. High concentrations affect the color and taste of beverages. Iron is not listed in the following tables, as there are often major differences between reported and actual concentrations. It may be added to water from well casings, pumps, and pipes. The concentration also is affected by micro-organisms. Special sampling and analytical techniques are needed for accurate study.
Manganese (Mn)	0.05 mg/l	Objectional for the same reason as iron. When both iron and manganese are present, it is recommended that the total concentration not exceed 0.3 mg/l. Micro-organisms also affect the concentration. Special techniques are needed for an accurate study.
Calcium (Ca) and Magnesium (Mg)		Principal causes for hardness and scale-forming properties of water. They reduce the lathering ability of soap.
Sodium (Na) and Potassium (K)		Impart a salty or brackish taste when combined with chloride. Sodium salts cause foaming in boilers.
Sulfate (SO_4)	250 mg/l	Commonly has a laxitive effect when the concentration is 600 to 1,000 mg/l, particularly when combined with magnesium or sodium. The effect is much less when combined with calcium. This laxitive effect is commonly noted by newcomers, but they become acclimated to the water in a short time. The effect is noticeable in almost all persons when concentrations exceed 750 mg/l. Sulfate combined with calcium forms a hard scale in boilers and water heaters.
Chloride (Cl)	250 mg/l	Large amounts combined with sodium impart a salty taste.
Fluoride (F)	2.0 mg/l	Concentrations of 0.8 to 1.3 mg/l are considered to play a part in the reduction of tooth decay. However, concentrations over 2.0 mg/l will cause the mottling of the enamel of children's teeth.
Nitrate (NO_3)	45 mg/l	Waters with high nitrate content should not be used for infant feeding as it may cause methemoglobinemia or cyanosis. High concentrations suggest organic pollution from sewate, decayed organic matter, nitrate in the soil, or chemical fertilizer. High nitrates in the natural waters of central Iowa are limited to isolated occurrences, usually from shallow dug wells on farms. Since the high concentrations are characteristic of individual wells and not of any one aquifer, nitrate will not be discussed in this report.
Dissolved Solids	500 mg/l	This refers to all of the material in water that is in solution. If affects the chemical and physical properties of water for many uses. Amounts over 2,000 mg/l will have a laxitive effect on most persons. Amounts up to 1,000 mg/l are generally considered acceptable for drinking purposes if no other water is available.
Hardness (as $CaCO_3$)		This affects the lathering ability of soap. It is generally produced by calcium and magnesium. Hardness is expressed in parts per million equivalent to $CaCO_3$ as if all the hardness were caused by this compound. Water becomes objectionable for domestic use when the hardness is above 100 mg/l; however, it can be treated readily by softening.
Temperature		Affects the desirability and economy of water use, especially for industrial cooling and air conditioning. Most users want a water with a low and constant temperature.
Suspended Sediment		Causes water to have a cloudy or muddy appearance. It must be settled or filtered out before the water is used. It is the material that "silts-up" reservoirs, and it is the major cause of the reduction of reservoir life.

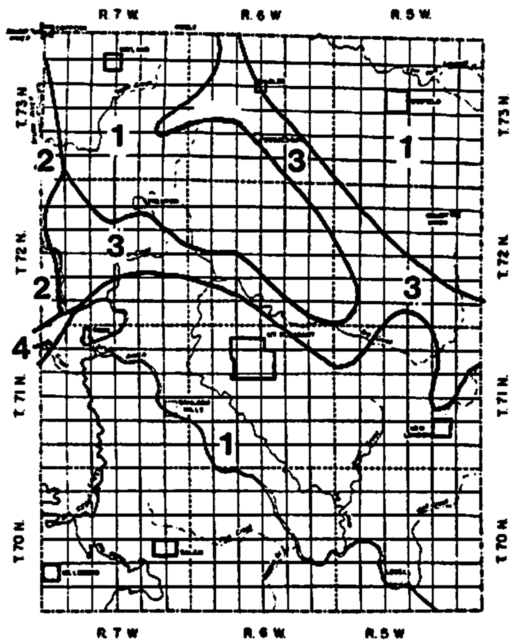
To the user, the quality of ground water is as important as the amount of water that an aquifer will yield. As ground water moves through soil and rock materials, it dissolves some of the minerals which, in turn, affect water quality. In addition to mineral content, bacterial and chemical contamination may be introduced through poorly constructed wells and seepage from other pollution sources.

Recommended standards for common water constituents are described in the table above. These are rationally accepted as guidelines for acceptable drinking water supplies. Limits for uses other than drinking often differ from these. For instance, water that is unacceptable for drinking and household use may be completely satisfactory for industrial cooling.

From analyses of ground water averages (A) and ranges (R) of values in milligrams per liter (mg/l) for several mineral constituents are summarized in Figures 13, 14 and 15 for the 4 major aquifers in Henry County. Recommended concentrations for some constituents are often exceeded without obvious ill effects, although the water may be unpalatable. Water quality analyses for individual wells should be obtained to determine if concentrations of constituents that affect health are exceeded.

Figure 13
CHEMICAL CHARACTER OF GROUND WATER

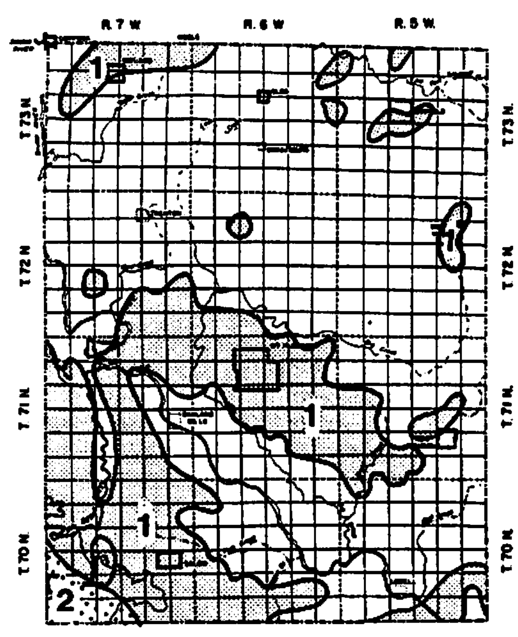
Surficial Aquifers



Area	Average and Range	Calcium (Ca)	Magnesium (Mg)	Sodium and Potassium (Na+K)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Fluoride (F)	Dissolved Solids	Hardness (as $CaCO_3$)
Drift Aquifer										
1	A	133	36	40	508	78	2.7	0.6	547	436
	R	77-150	24-61	13-83	304-739	11-331	0.5-7.5	0.2-0.7	357-932	292-625
Alluvial Aquifer										
2	A	78	23	18	248	99	14	.3	415	290
	R	59-91	11-31	7.6-37	168-327	58-163	3-25	.2-.4	305-535	216-360
Buried Channel Aquifer										
3	A	53	20	153	579	46	12	.4	591	215
	R	24-88	3.1-39	87-244	410-803	.2-170	2.5-28	0-1.0	357-715	78-360
4	A	160	57	95	676	250	2.0	.4	946	616
	R	128-175	46-61	67-130	548-775	110-362	.5-4.0	.2-.9	782-1120	510-685

Surficial aquifers yield the least mineralized water and of best quality of all ground water sources in Henry County. Water from the alluvial and drift aquifers are of good quality and found in large amounts. The dissolved solids content tends to be a bit high, but is acceptable for drinking purposes if no other water is available. The water from the buried-channel aquifer is often more mineralized than water from the other two surficial aquifers, because they are next to the bedrock, through which they have cut their channels. Their dissolved solids content is higher than the other surficial aquifers. Water temperatures average 54°F (12°C) and the range of temperatures is from 48°F to 58°F (9.0°C to 14.5°C).

Mississippian Aquifer
Upper Part



Aquifer not present

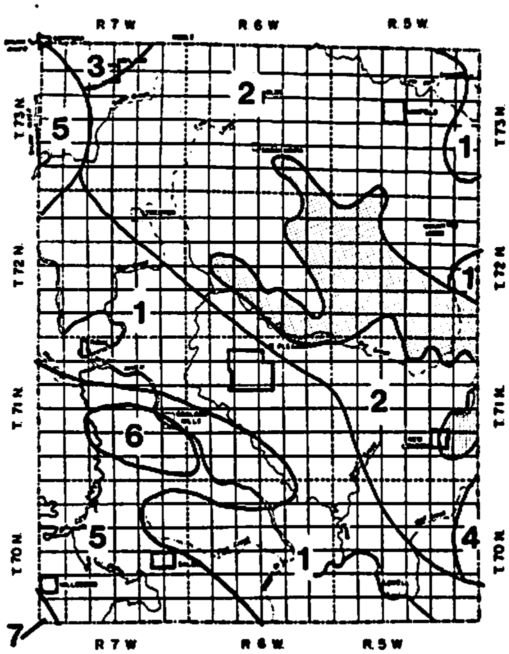
Area	Average and Range	Calcium (Ca)	Magnesium (Mg)	Sodium and Potassium (Na+K)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Fluoride (F)	Dissolved Solids	Hardness (as $CaCO_3$)
Upper Part of the Mississippian Aquifer										
1	A	104	32	43	472	91	4.5	0.4	537	399
	R	66-156	19-50	6.8-118	322-802	5.8-240	0.5-24	0-1.8	345-737	315-545
2	A	299	92	125	281	1100	8.5	.8	1950	1130
	R	197-547	47-124	33-214	44-434	750-1590	2-22	.3-1.6	1420-2740	723-1560

Good to fair water quality is available in the Upper Part of the Mississippian in area 1. The water is more highly mineralized than that typically found in the surficial aquifers and usually hard. The dissolved solids content is just slightly over recommended levels. In area 2, with the dissolved solids and sulfate content being greatly exceeded, the quality of the water greatly decreases. Average water temperature is 55°F (13°C) and the range of temperatures is from 51°F to 60°F (10.5°C to 15.5°C).

Figure 14

Mississippian Aquifer

Lower Part



Aquifer not present

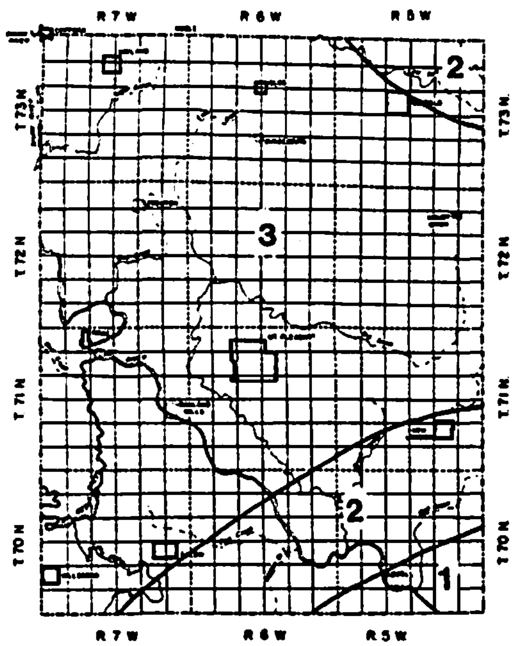
Area	Average and Range	Calcium (Ca)	Magnesium (Mg)	Sodium and Potassium (Na+K)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Fluoride (F)	Dissolved Solids	Hardness (as $CaCO_3$)
Lower Part of the Mississippian Aquifer										
1	A	90	37	42	504	46	6	0.4	490	380
	R	30-160	10-61	9-107	293-710	1-186	0-69	0-1.2	280-800	160-575
2	A	42	20	117	453	57	14	.9	477	190
	R	22-70	12-35	55-195	344-551	1-160	.5-57	0.5-1.5	281-677	122-317
3	A	122	43	61	344	309	6	.6	794	482
	R	92-152	40-46	48-74	335-351	258-360	4.5-7	.4-.7	728-850	394-570
4	A	140	68	318	440	850	55	.8	1730	630
5	A	78	38	276	592	354	57	1.6	1110	355
	R	39-128	19-68	143-489	465-754	260-560	.5-150	.5-2.5	879-1480	176-561
6	A	102	52	718	459	1340	169	2.6	2710	469
	R	35-193	15-107	451-994	266-595	920-1660	19-355	.5-6	2220-3250	148-891
7	A	215	82	1375	956	2520	610	1.8	5260	875
	R	102-452	58-120	1040-1630	378-3640	1800-4020	290-930	1.4-3	4070-7330	497-1623

Water in the lower part of the Mississippian aquifer is generally of poorer quality than found in the upper part. Throughout the county the water is exceptionally hard. Areas 3 through 7 have very high dissolved solids content. Areas 3, 4 and 5 have sulfate concentrations which slightly exceed the recommended limit, while areas 6 and 7 in the southwestern corner, have sulfate concentrations between 1,000 and 2,500 mg/l which greatly exceeds the standard. Area 7 has a high chlorine content, with a marginal fluoride content, while area 6's fluoride content of 2.6 exceeds the standards at 2.0. Areas 1 and 2 seem to have the best quality water of the lower Mississippian aquifers. Average water temperature is 55°F (13°C), and the range of temperature from 51°F to 60°F (10.5°C to 15.5°C).

Figure 15

CHEMICAL CHARACTER OF GROUND WATER

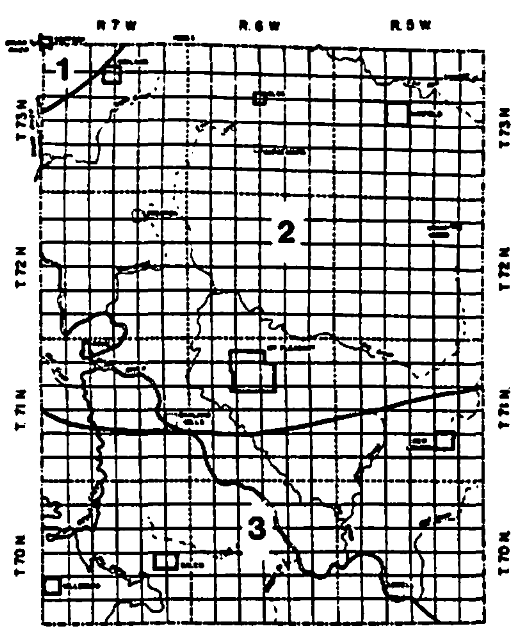
Devonian Aquifer



Area	Average and Range	Calcium (Ca)	Magnesium (Mg)	Sodium and Potassium (Na+K)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Fluoride (F)	Dissolved Solids	Hardness (as $CaCO_3$)
1	A	333	111	485	282	1770	218	2.1	3310	1290
	R	246-378	79-146	412-539	244-305	1470-2070	209-235	1.0-3.0	2800-3800	942-1540
2	A	359	116	1500	342	3450	516	2.4	6340	1380
	R	180-532	75-157	1330-1630	303-383	2200-4000	400-810	1.7-4.2	5050-6900	760-1870
3	A	492	138	2220	289	4580	1100	2.8	9570	1800
	R	441-617	97-155	1850-2700	183-378	4030-5000	550-1740	1.9-5.0	8240-11,100	1580-2180

The Devonian aquifer in this county possesses very highly mineralized water and is found to be of very poor quality. The water is highly mineralized with standard exceeding concentrations of sodium, sulfate, chloride, and fluoride. The dissolved solids content range from 3,000 to 10,000 mg/l. Water temperatures are higher than that from the Mississippian aquifers sources averaging 60°F (15.5°C) and with a temperature range of 54°F to 64°F (12.0°C to 18.0°C).

Cambro-Ordovician Aquifer



Area	Average and Range	Calcium (Ca)	Magnesium (Mg)	Sodium and Potassium (Na+K)	Bicarbonate (HCO_3)	Sulfate (SO_4)	Chloride (Cl)	Fluoride (F)	Dissolved Solids	Hardness (as $CaCO_3$)
1	A	106	50	202	304	552	52	1.2	1180	470
	R	98-116	46-54	192-211	283-337	520-600	38-60	1.0-1.5	1120-1240	452-510
2	A	93	42	232	295	520	79	1.4	1130	406
	R	86-105	37-47	223-249	283-305	489-543	69-85	1.2-1.6	1110-1150	372-455
3	A	84	34	267	298	476	124	1.7	1160	349
	R	78-92	26-41	247-283	283-317	455-500	100-148	1.0-2.2	1100-1220	322-388

This deep aquifer yields water of relatively good quality compared to the other rock aquifers. The water is noticeably hard and just slightly exceeds recommended standards for sulfate and has dissolved solids concentrations of about 1,000 mg/l, but it is not as highly mineralized as that from parts of the Mississippian and Devonian Aquifers. Water temperatures are higher than other rock aquifer sources averaging 72°F (22°C) and with a temperature range from 68°F to 76°F (20.0°C to 24.5°C).

RECOMMENDATIONS FOR PRIVATE WATER WELLS

Contracting for Well Construction

To protect your investment and guarantee satisfactory well completion, it is a good idea to have a written agreement with the well driller. The agreement should specify in detail:

- size of well, casing specifications, and types of screen and well seal
- methods of eliminating surface and subsurface contamination
- disinfection procedures to be used
- type of well development if necessary
- test pumping procedure to be used
- date for completion
- itemized cost list including charges for drilling per foot, for materials per unit, and for other operations such as developing and test pumping
- guarantee of materials, workmanship, and that all work will comply with current recommended methods
- liability insurance for owner and driller

Well Location

A well should be located where it will be least subject to contamination from nearby sources of pollution. The Iowa State Department of Health recommends minimum distances between a new well and pollution sources, such as cesspools (150 ft.), septic tanks (50 ft.), and barnyards (50-100 ft. and downslope from well). Greater distances should be provided where possible.

The well location should not be subject to flooding or surface water contamination. Select a well-drained site, extend the well casing a few feet above the ground, and mound earth around it. Diversion terraces or ditches may be necessary on slopes above a well to divert surface runoff around the well site.

In the construction of all wells care should be taken to seal or grout the area between the well bore and the well casing (the annulus) as appropriate so that surface water and other pollutants cannot seep into the well and contaminate the aquifer.

Locate a well where it will be accessible for maintenance, inspection, and repairs. If a pump house is located some distance from major buildings and wired separately for power, continued use of the water supply will be jeopardized by fire in major buildings.

Water Treatment

Water taken from a private well should ideally be tested every six months. The University Hygienic Laboratory will do tests for coliform bacteria, nitrate, iron, hardness, and iron bacteria in drinking water for private individuals. Special bottles must be used for collecting and sending water samples to the laboratory. A sample kit can be obtained by writing to the University Hygienic Laboratory, University of Iowa, Oakdale Campus, Iowa City, Iowa 52242. Indicate whether your water has been treated with chlorine, iodine, or bromine; for different sample bottles must be used for treated and untreated water. The charge for the bacterial test is $3.00; for iron hardness and nitrate, it is $3.00; and for iron bacteria, $5.00. If your well is determined to be unsafe, advice for correcting the problem can be obtained from your county or state Department of Health. Several certified private laboratories also run water analyses.

Shock chlorination is recommended following the construction and installation of a well and distribution system and anytime these are opened for repairs or remodeling a strong chlorine solution is placed in the well and complete distribution system to kill nuisance and disease-causing organisms. If the first shock chlorination does not rid the water supply of bacteria it should be repeated, if this does not solve the problem the well should be abandoned or the water should be continuously disinfected with proper chlorination equipment.

Since most of the ground waters in Henry County are mineralized, water softening and iron removal equipment may make water more palatable and pleasant to use. Softened water contains increased sodium; contact your physician before using a softener if you are on a sodium-restricted diet. Chlorination followed by filtration will remove most forms of iron and iron bacteria. Iron bacteria has no adverse effect on health but will plug wells, water lines, and equipment and cause tastes and odors. Iron removal equipment can be used if problems persist.

Well Abandonment

Wells taken out of service provide easy access for pollution to enter aquifers supplying water to other wells in the vicinity. Unprotected wells may also cause personal injury. Proper abandonment procedures should be followed to restore the natural conditions that existed before well construction and prevent any future contamination. Permanent abandonment requires careful sealing. The well should be filled with concrete, cement grout, or sealing clays throughout its entire length. Before dug or bored wells are filled at least the top 10 feet of lining should be removed so surface waters will not penetrate the subsurface through a porous lining or follow cracks in or around the lining. The site should be completely filled and mounded with compacted earth.

ABANDONED WELLS SHOULD NEVER BE USED FOR DISPOSAL OR SEWAGE OR OTHER WASTES.

SOURCES OF ADDITIONAL INFORMATION

In planning the development of a ground water supply or contracting for the drilling of a new well additional or more specific information is often required. This report section lists several sources and types of additional information.

State Agencies That May Be Consulted

Iowa Geological Survey[1]	123 North Capitol Iowa City 52242	(319) 338-1173
State Health Department[2,6]	Lucas Building Des Moines 50319	(515) 281-5787
Iowa Natural Resources Council[3]	Wallace Building Des Moines 50319	(515) 281-5914
Iowa Dept. of Environ. Quality[4]	Wallace Building Des Moines 50319	(515) 281-8854
University Hygienic Laboratory[5]	U. of IA, Oakdale Campus Iowa City 52242	(319) 353-5990
Cooperative Extension Service in[6] Agriculture and Home Economics	110 Curtis Hall, ISU Ames 50011	(515) 294-4569

Functions:

[1] Geologic and ground water data repository, consultant on well problems, water development and related services

[2] Drinking water quality, public and private water supplies

[3] Water withdrawal regulation and Water Permits for wells withdrawing more than 5000 gpd

[4] Municipal supply regulation and well construction permits

[5] Water quality analysis

[6] Advice on water systems design and maintenance

Well Drillers and Contractors

The listing provided here was drawn from an Iowa Geological Survey mailing list and yellow pages of major towns in phone books. These selected are within an approximate radius of 50 miles of Henry County. For a statewide listing contact either the Iowa Water Well Drillers Association, 4350 Hopewell Ave., Bettendorf, Iowa 51712, (319) 355-7528 or the Iowa Geological Survey, (319) 338-1173.

Bailey Well Co.
203 E. Main
New London, Iowa 52645

Dwayne Bruinekool
Bruinekool Well Co.
Oskaloosa, Iowa 52577

Detrick Well Co.
R. R. # 1
New London, Iowa 52645

Gingerich Well Co.
Kalona, Iowa 52247

Jack Kramer
Mt. Pleasant, Iowa 52641

Latta and Sons Well Drilling
Riverside, Iowa 52327

Duane Latta
Latta Well and Pump
Rural Route
Wilton, Iowa 52778

Neal Lyon Well Co.
Salem, Iowa 52649

Miller and Son Well Co.
Kalona, Iowa 52247

Novotny Well Co.
Shueyville, Iowa 52338

Schlicher Brothers Well Co.
Hwy. 34 West
Fairfield, Iowa 52556

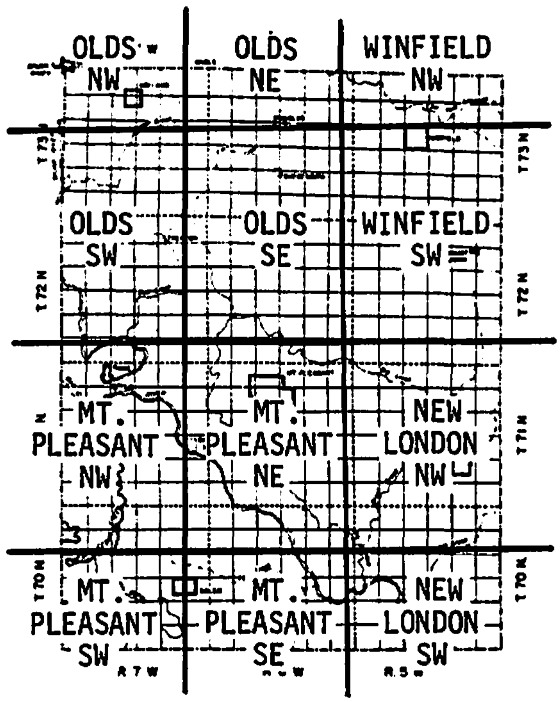
Schlicher Well Co.
P. O. Box 207
Donnelson, Iowa 52625

Schmeiser Well Co.
1111 Hageman St.
Burlington, Iowa 52601

Wayne Smith
Box 195
West Liberty, Iowa 52776

Wilson Well Co.
R. R. # 3
Burlington, Iowa 52601

Topographic Maps (Available from the Iowa Geological Survey)



Map Title	Date	Scale	Contour Interval
Olds NW	(Preliminary) anticipated 7½-	1:24,000	10'
Olds NE	minute quadrangle publication 12/80	1:24,000	10'
Winfield NW		1:24,000	10'
Olds SW		1:24,000	10'
Olds SE		1:24,000	10'
Winfield SW		1:24,000	10'
Mt. Pleasant NW		1:24,000	10'
Mt. Pleasant NE		1:24,000	10'
New London		1:24,000	10'
Mt. Pleasant SW		1:24,000	10'
Mt. Pleasant SE		1:24,000	10'
New London SW		1:24,000	10'

Useful Reference Materials

Coble, R.W., and Roberts, J.V., 1971, The water resources of Southeast Iowa, Iowa Geological Survey, Water Atlas No. 4.

Horick, P.J., and Steinhilber, W.L., 1973, Mississippian aquifer of Iowa, Iowa Geological Survey, Misc. Map Series No. 3.

Horick, P.J., and Steinhilber, W.L., 1978, Jordan aquifer of Iowa, Iowa Geological Survey, Misc. Map Series No. 6.

Iowa State Department of Health, 1971 Sanitary standards for water wells, State Department of Health, Environmental Engineering Service.

Van Eck, O.J, 1971, Optimal well plugging procedures, Iowa Geological Survey, Public Information Circular No. 1.

Van Eck, O.J, 1978, Plugging procedures for domestic wells, Iowa Geological Survey, Public Information Circular No. 11.